Molekularer Mechanismus und strukturelle Implikationen des $pH\text{-}Gatings\ einwärtsgleichrichtender}\ K^{\scriptscriptstyle +}\text{-}Kan\"{a}le\ (K_{\rm ir})$

pH-Gating of Inward-Rectifier K^+ Channels (K_{ir}) : Molecular Mechanism and Structural Implications

Dissertation

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Abbreviations

Throughout the text, the single-letter code for amino acids is used.

Connotation of point mutations: (K80M) = K (lysine) at position 80 replaced by M (methionine)

ABC proteins ATP-binding cassette proteins

aBS antenatal Bartter Syndrome

CFTR cystic fibrosis transmembrane conductance regulator, a chloride

channel of the ABC protein family

CNG cyclic nucleotide gated channel

cRNA mRNA synthesized *in-vitro* from complementary DNA

Cu-Phen Cu(II)-1,10-phenanthroline

DEPC diethylpyrocarbonate

DTNB 5,5'-dithio-bis[-2-nitrobenzoic acid]

DTT dithiothreitol

EAG ether-à-go-go-related gene encoded K⁺ channel

 E_{K^+} equilibrium potential for the ion K^+ according to the Nernst equation

ENac epithelial Na⁺ channel

EPR electron paramagnetic resonance

FmocCl 9-fluorenyl-methoxycarbonyl chloride

GYG glycine-tyrosine-glycine; highly conserved motif termed selectivity

filter of K+ channels

h human

I-V current-voltage relation

 K_{ATP} ATP sensitive K^+ channel

KcsA prokaryotic K⁺ channel from *Streptomyces lividans*

K_{int} solution applied to the cytoplasmic side of an inside-out patch

(composition see "Materials and Methods")

 K_{ir} inward-rectifier K^+ channel

KQT "long-QT" K⁺ channel; "long-QT" refers to a prolonged repolarization

period (Q-T interval) of cardiac action potentials measured in electro-

cardiograms

K_v superfamily of voltage-gated K⁺ channels

m mouse

MTSES (2-sulfonatoethyl)-methanethiosulfonate

NMDG N-methyl-d-glucosamine

PGE₂ prostaglandin E₂

pH_i intracellular pH

PIP2 phosphatidylinositol-4,5-bisphosphate

pK_{app} apparent pK

PVP polyvinyl-pyrrolidone

r rat

RKR triad structural arrangement, in which two arginine and a lysine residue

closely interact

ROMK renal outer medulla K⁺ channel

SK small conductance K⁺ channel

Slo slopoke-like K⁺ channels, also termed BK-type K⁺ channels

SPM spermine

SSCP single-strand conformation polymorphism

TOK two P-domain outward-rectifier K⁺ channel

TWIK two P-domain weak inward-rectifier K⁺ channel

V_m membrane potential

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1. Introduction

1.1 Inward-rectifier K⁺ channels

1.1.1 Superfamily of P-loop channel proteins

Ion channels are integral membrane proteins that are used in all types of cells for transport of ions across plasma membranes. In the early ages of electrophysiology these channels have been characterized by their ion selectivity, conductance and time dependent activation and deactivation properties (termed gating). With the molecular cloning of many genes coding for ion selective channels during the last 15 years, a classification has been based on the primary structure of the channel proteins (Wei et al., 1996):

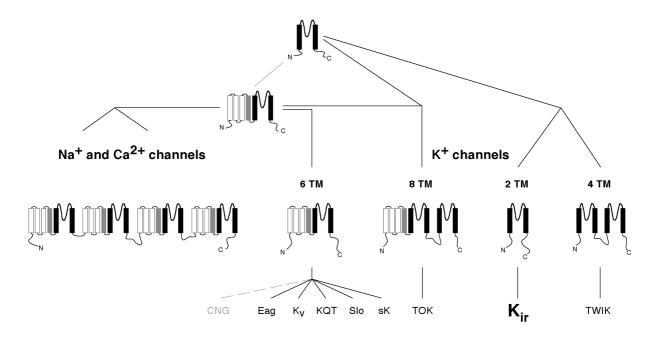


Fig. 1. Superfamily of mammalian P-loop channel proteins forming cation selective channels. Transmembrane domains black: conserved core M1 (S5) and M2 (S6); grey: S4-helix (voltage sensor in voltage-gated channels); white: S1, S2, S3. N- and C-termini are located intracellularly.

Ion channel families (grouped according to the number of transmembrane domains TM): CNG (cyclic nucleotide-gated channels), EAG (ether-à-go-go-related channels), $K_{\rm V}$ (voltage-gated K^+ channels), KQT (long-QT K^+ channels), Slo (BK-type K^+ channels), sK (small conductance K^+ channels), TOK (two P-domain outward-rectifier K^+ channels), $K_{\rm ir}$ (inward-rectifier K^+ channels) and TWIK (two P-domain weak inward-rectifier K^+ channels).

Na $^+$ -, K $^+$ - and Ca $^{2+}$ -channels are structurally related and seem to have originated from a common ancestor early in evolution. The conserved core region consists of two membrane spanning domains (commonly designated M1 and M2) flanking a pore-forming loop. This P-loop (or H5-region) is highly conserved and defines the superfamily of P-loop channel proteins. Later on, the structural motif of M1-P-M2 has been extended by four additional trans-membrane segments (S1-S5-P-S6), and subsequent fusion and diversification of these genes created the variety of cation selective ion channel subunits known so far. As illustrated in Fig. 1, inward-rectifier K^+ channels (K_{ir}) are the smallest proteins in this superfamily. Their simple membrane topology resembles the ancestral channel architecture (see Fig. 1, top) which is also found in bacterial homologs (Schrempf et al., 1995).

1.1.2 Molecular structure

The membrane topology and secondary structure of K_{ir} channels has been predicted from hydropathy plots of the primary sequence and investigated by detailed mutagenesis. Two hydrophobic segments M1 and M2 form membrane spanning domains and flank a stretch of amino acids that is highly homologous to the H5- or P-region. N- and C-termini are thought to be located in the cytoplasm, although a partial association with the membrane can not be excluded. There is experimental evidence that parts of the C-terminus contribute to the intracellular vestibule of the pore (Taglialatela et al., 1994; Baukrowitz et al., 1999).

Recently, the crystal structure of the core region of KcsA, a K^+ channel from the bacterium *Streptomyces lividans* with structural homology to K_{ir} channels, has been resolved with a resolution of 3.2 Å (Doyle et al., 1998). The results are in good agreement with structural predictions and provide a basis for understanding ion selectivity and permeation. The N-terminal part of the P-loop is -helical and forms the shallow outer vestibule of the channel. Ion selectivity is determined by a narrow binding site in the pore termed selectivity filter which is formed by the backbone carbonyls of the GYG consensus motif. The long and narrow inner vestibule is formed by residues of the second transmembrane helix (M2) and contains binding sites for K^+ and small blocking ions. M1 also lines the pore but is tilted around M2. However, the structure of intracellular domains (N- and C-terminus) remains unknown.

In analogy to voltage gated K^+ channels (MacKinnon, 1991), K_{ir} channels are assembled from four subunits. This was shown by a functional approach (Glowatzki et al., 1995) as well as biochemically (Yang et al., 1995b). Most K_{ir} subunits form functional homotetramers, some require co-assembly with other K_{ir} subunits in order to form functional channels (Fakler et al., 1996a). Formation of heteromeric channels seems to underly several K_{ir} conductances of physiological importance. For example $K_{ir}3.1/K_{ir}3.4$ heterotetrameric channels closely resemble the native atrial $I_{K,ACh}$ (Krapivinsky et al., 1995b).

1.1.3 Physiological functions

Inward-rectifier potassium channels maintain the membrane potential (V_m) near the K⁺ reversal potential (E_{K^+}) , in excitable and non-excitable cells. These channels mediate a high K^+ conductance at E_{K^+} and at voltages slightly positive to E_{K^+} , which decreases when the membrane is further depolarized (Hille, 1992). Thus inward-rectifier K⁺ channels have a highly stabilizing effect on V_m around E_{K^+} but allow depolarization of the cell. The range of membrane potential over which inward-rectifier K+ channels stabilize Vm basically depends on their voltage-dependence of rectification which may be "strong" or "mild" (Hille, 1992). Strong inward-rectifier K⁺ channels have been described in a wide variety of cells including skeletal muscle (Leech and Stanfield, 1981), cardiac muscle cells (Kurachi, 1985; Vandenberg, 1987; Matsuda, 1991), starfish egg cells (Hagiwara et al., 1976), endothelial cells (Silver and DeCoursey, 1990) and osteoclasts (Kelly et al., 1992). In the current-voltage relation (I-V) of strong inward-rectifier channels the outward current exhibits a maximum at potentials close to E_{κ^+} which is followed by a region of negative slope conductance at more positive potentials (see Fig. 2 below). This outward current maximum is physiologically important since it determines a "trigger threshold" of excitation: whenever a depolarizing current exceeds the maximal outward current, the inward-rectifier K^+ channels close down and V_m is free to change (Hille, 1992).

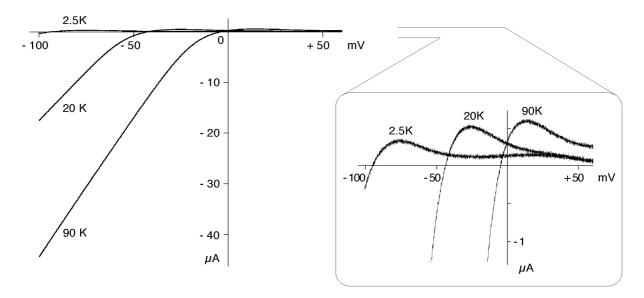


Fig. 2. Strong inward-rectification of K^+ currents recorded from *Xenopus* oocytes injected with $K_{ir}2.1$ cRNA in a two-electrode-voltage-clamp experiment. Note the shift in reversal potential E_{K^+} and slope conductance for 2.5, 20 and 90 mM extracellular K^+ . Inset: enlarged current scale, showing the change in conductance around the reversal potential.

In contrast to that, weak inward-rectifiers conduct significant outward currents over the whole physiological voltage range (for an example see Fig. 12). This has two consequences: they suppress electrical excitation and they are able to effectively secrete K⁺ from cells. Inhibition of

electrical excitation is an important mechanism to protect muscle cells during metabolic exhaustion or to regulate insulin secretion in pancreatic -cells (Ashcroft, 1988). A decrease of intracellular ATP levels activates K_{ir} channels of the K_{ATP} -type (see section 3.1.5) and the resulting hyperpolarisation reduces the energy demand of the cell. In the kidney, K^+ secretion by epithelial cells is crucial for K^+ homeostasis and NaCl reabsorption (Lang and Rehwald, 1992; Wang et al., 1992; Wang, 1995). Active transport processes provide the concentration gradient for K^+ efflux through weak inward-rectifier K^+ channels (see section 1.2.3).

1.1.4 Mechanism and determinants of inward-rectification

The hallmark property of K_{ir} channels is their inward-rectification of K^+ -currents which depends on both, membrane voltage (V_m) and K^+ equilibrium potential E_{K^+} (Hille, 1992). These parameters determine the driving force for K^+ ($V_d = V_m - E_{K^+}$), which directs K^+ flow. As the molecular mechanism underlying the phenomenon of inward-rectification, block of K_{ir} channels by intracellular Mg^{2^+} and the polyamines spermine and spermidine has been identified (Vandenberg, 1987; Fakler et al., 1994b, 1995; Ficker et al., 1994; Lopatin et al., 1994). The pore-block by these intracellular cations is characterized by a voltage-dependence which may be strong ('strong rectifiers') or weak ('weak rectifiers') and which is determined by the quantity of V_m - E_{K^+} rather than by E_m alone (Hagiwara et al., 1976; Leech and Stanfield, 1981; Cohen et al., 1989; Hille, 1992). Since the latter was primarily found for changes in extracellular K^+ concentration ($[K^+]_{ex}$) inward-rectification is more correctly said to depend on $[K^+]_{ex}$ and V_m but not on the intracellular K^+ concentration ($[K^+]_{in}$). To account for this fact, an extracellular binding site for K^+ ions on K_{ir} channels was hypothesized (Hille, 1992).

Comparison of the primary sequence of strongly and weakly rectifying K_{ir} channels identified two structural determinants involved in inward-rectification. These are negatively charged residues, one in the second transmembrane segment (M2-site) (Fakler et al., 1994b; Lu and MacKinnon, 1994; Stanfield et al., 1994; Wible et al., 1994), the other in the cytoplasmic C-terminal domain (C-terminal-site) (Taglialatela et al., 1995; Yang et al., 1995a). Inward-rectifiers of the $K_{ir}2$ subfamily that exhibit both the M2- and C-terminal-site display complex kinetics of spermine block (Lopatin et al., 1995; Fakler and Ruppersberg, 1996). In contrast, those carrying only the M2-site show monoexponential blocking behavior and their steady state block is described by a single Boltzmann function (Fakler et al., 1994b; Glowatzki et al., 1995). Voltage-dependence of block is usually quantified in terms of the change in membrane voltage necessary for an e-fold increase in block. This parameter is assumed to correlate with the number of blocking charges times the percentage of the transmembrane electric field which these charges move through in the blocking reaction (electrical distance according to Woodhull (1973)).

In agreement with these determinants, all K_{ir} channels known to date can be classified as either strong or weak rectifiers, except for $K_{ir}6.2$. Channels formed from this subunit are unique in that polyamine-mediated rectification is not fixed but changes with intracellular pH in the physiological range (Baukrowitz et al., 1999). Inward-rectification is prominent at basic pH, while at acidic pH rectification is very weak. Mutagenesis uncovered a titratable C-terminal histidine residue (H216) in $K_{ir}6.2$ as the structural determinant and electrostatic interaction between this residue and spermine as the molecular mechanism underlying pH-dependent inward-rectification.

1.1.5 The family of inward-rectifier K⁺ channels

In 1993, the first cDNA clones encoding inward-rectifier K^+ channels - termed ROMK ($K_{ir}1.1$, from kidney (Ho et al., 1993)) and IRK1 ($K_{ir}2.1$, from a mouse macrophage library (Kubo et al., 1993)) - were successfully isolated by expression cloning. Until today, cloning based on sequence homology resulted in discovery of more than a dozen of K_{ir} genes (Fig. 3). They all code for proteins of 361-502 amino acids length with the characteristic membrane topology described in section 1.1.2. The short intracellular N-terminus comprises about 80 amino acids in all K_{ir} subunits whereas the C-terminus varies in length between 201 and 315 amino acids.

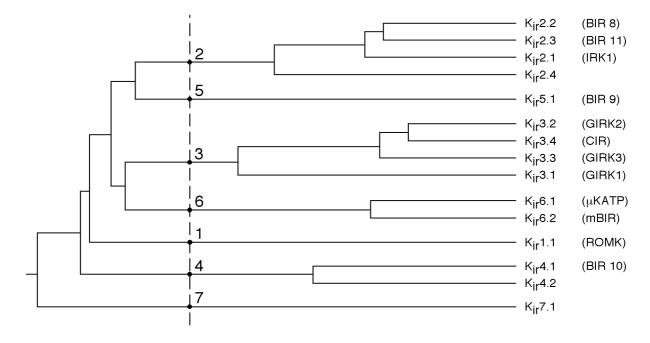


Fig. 3. The family of mammalian inward-rectifier K⁺ channels (K_{ir}). Primary sequences of the K_{ir} subunits indicated were aligned using the Clustal method (Hein, 1989) with the PAM250 residue weight table. Distances in the tree are based on the homology index calculated by the program. The dashed line defines seven different K_{ir} subfamilies. The generic nomenclature is given together with the original names published (in parentheses).

A first classification based on primary sequence homology (Doupnik et al., 1995) proposed 6 different subfamilies. Although this nomenclature is widely accepted, it poorly reflects functional homologies. Other reviews proposed four K_{ir} subgroups (including new genes) based on both functional and structural properties (Fakler and Ruppersberg, 1996; Nichols and Lopatin, 1997).

The weak inward-rectifier K_{ir}1.1 (ROMK1) displays high sensitivity to changes in intracellular pH and is primarily expressed in kidney and uterus (Ho et al., 1993). K_{ir}6 (µK_{ATP}, (Inagaki et al., 1995b) and mBIR (Inagaki et al., 1995a)), another subfamily of weak inward-rectifiers, forms ATP-sensitive channels (K_{ATP}) by assembly with sulfonylurea receptors (SUR1 (Aguilar-Bryan et al., 1995), SUR2A (Inagaki et al., 1996) or SUR2B (Isomoto et al., 1996)), members of the superfamily of ATP-binding cassette proteins. They are expressed in a wide variety of tissues including pancreatic -cells, cardiac myocytes, skeletal muscle and brain. The K_{ir}4 subfamily comprises two strong inward-rectifiers expressed in brain (BIR10 (Bond et al., 1994) and kidney (K_{ir}4.2 (Shuck et al., 1997)). Members of the K_{ir}3 subfamily are strong rectifiers primarily expressed in brain and heart (Dascal et al., 1993; Ashford et al., 1994; Lesage et al., 1994). A hallmark property is their activation by direct binding of ("G-protein coupling" (Huang et al., 1995; Krapivinsky et al., 1995a). Subfamily K₁₁2 comprises strong inward-rectifiers with complex blocking characteristics expressed in a variety of cells (K_i,2.1 (Kubo et al., 1993), K_i,2.2 (Takahashi et al., 1994), K_i,2.3 (Bond et al., 1994) and K_{ii}2.4 (Krapivinsky et al., 1998)). The other subfamilies are poorly characterized since K_{ir}5.1 (BIR9 (Bond et al., 1994)) does not express homomeric channels and K_{ir}7.1 has only recently been cloned (Partiseti et al., 1998).

1.2 K_{ir}1.1 channels (ROMK)

1.2.1 romk splice variants

 $K_{ir}1.1$ was originally cloned from a rat kidney cDNA library as the first member of the K_{ir} family. Genomic analysis revealed, that human and rat *romk* genes contain 6 exons, which are spliced alternatively to yield the three isoforms (Shuck et al., 1994; Yano et al., 1994; Boim et al., 1995; Kondo et al., 1996) as depicted in Fig. 4. Since exon 6 (core) encodes the major part of the channel protein, splicing results primarily in variable length of the N-terminus. rROMK2 ($K_{ir}1.1b$) is shortened by 19 amino acids, and rROMK3 ($K_{ir}1.1c$) is extended by 7 amino acids compared to rROMK1 ($K_{ir}1.1a$).

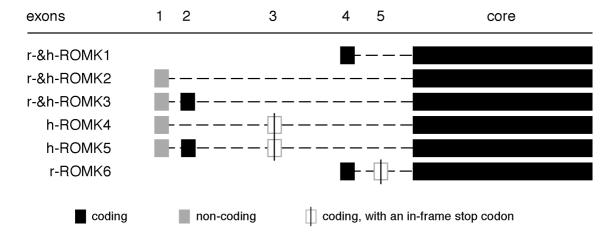


Fig. 4. Alternative splicing of the *romk* gene (adapted from Kondo et al. (1996)) yielding 6 different ROMK mRNAs. Exons 2 and 4 code for different N-termini in ROMK3 (K_{ir}1.1c) and ROMK1 (K_{ir}1.1a), respectively. Exons 1, 3 and 5 do not contribute to the primary structure of ROMK subunits. Exon 6 (core) encodes ROMK2 (K_{ir}1.1b) and the major part of the other K_{ir}1.1 splice variants.

ROMK1 as well as the other splice variants are differentially expressed in renal tubular cells (Boim et al., 1995; Lee and Hebert, 1995), except rROMK6 mRNA which is found in several other tissues (Kondo et al., 1996). Up to date, no functional differences could be observed between $K_{ir}1.1$ isoforms when expressed in *Xenopus* oocytes. Thus, all $K_{ir}1.1$ experiments in this study were carried out with the $rK_{ir}1.1$ a splice variant unless stated otherwise.

1.2.2 Regulation of channel activity

Tsai and colleagues showed that $K_{ir}1.1$ channels are particularly sensitive to changes in intracellular pH (Tsai et al., 1995). pH_i did not affect the single channel amplitude, instead a decrease in channel open probability was observed upon acidification (Fakler et al., 1996, Choe et al., 1997, McNicholas et al., 1998). The steady-state current-pH_i relation showed a pH_i value for half maximal activation (pK_{app}) of 6.9 and a Hill coefficient of around 3 indicating cooperativity of the gating process. Binding of K⁺ ions to an extracellular site has been shown to be essential for $K_{ir}1.1$ channel activity but not for that of other K_{ir} channels (Doi et al., 1996). The halfmaximal K⁺ concentration determined in whole-cell experiments was around 4.5 mM but shifted to higher values when the intracellular pH was decreased.

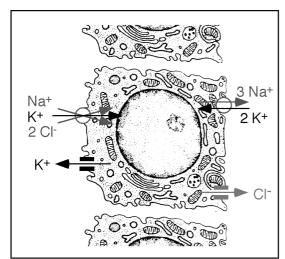
Binding of the negatively charged phospholipid PIP2 appears to be a general property of K_{ir} channels (Hilgemann and Ball, 1996; Huang et al., 1998), and has also been reported for $K_{ir}1.1$. Fusion proteins constructed from the $K_{ir}1.1$ C-terminus bound PIP2 *in-vitro*, and the presence of PIP2 was shown to be necessary for $K_{ir}1.1$ channel activity (Huang et al., 1998). Finally, regulatory phosphorylation by PKA has been demonstrated for $K_{ir}1.1$ channels in renal epithelial cells (McNicholas et al., 1994) and biochemically verified in detail (Xu et al., 1996).

1.2.3 Physiological role in the kidney

Intracellular K^+ (approximately 145 mM) represents the major portion of total body K^+ . The K^+ concentration in extracellular fluids ranges from 3.5-5 mM. To maintain a constant serum K^+ level, 95% of dietary K^+ absorbed from the intestine is excreted through the kidney and the remaining portion is eliminated via the colon (Thier, 1986; Stanton, 1989). Under pathophysiological conditions like chronic renal failure, colonic excretion is increased and can contribute significantly to K^+ homeostasis (Martin et al., 1986).

K⁺ secretion in the kidney is a very complex process depending on flow rate, luminal K⁺, Na⁺ and Cl⁻ concentrations, hormones and the acid-base status (Stanton, 1989; Wang, 1995; Giebisch, 1998).





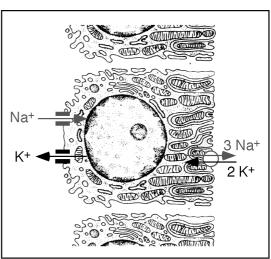


Fig. 5. (A) Salt reabsorption in tubular cells of the ascending loop of Henle; left (luminal side): apical membrane with Na⁺/K⁺/2Cl⁻ cotransporter and K_{ir}1.1 channels; right: basolateral membrane with Na⁺/K⁺-ATPase and cAMP-dependent Cl⁻ channels.

(B) K⁺ secretion in principal cells in distal tubule and cortical collecting duct; left (luminal side): apical membrane with K_{ir}1.1 channels and epithelial Na⁺ channels (ENaC); right: basolateral membrane with Na⁺/K⁺-ATPase. Arrows indicate the direction of ion flux under physiological conditions.

In the ascending loop of Henle, K⁺ participates in the reabsorption of NaCl from the primary urine as illustrated in Fig. 5 A. Luminal Na⁺, K⁺ and Cl⁻ enters the tubular cells via the furosemide-sensitive Na⁺/K⁺/2Cl⁻ cotransporter. This process depends on K⁺ efflux through apical K⁺ channels (K_{ir}1.1), which allow recycling of luminal K⁺ (Hebert, 1998). Na⁺ and Cl⁻ are eliminated from the cells by cAMP-dependent Cl⁻ channels and Na⁺/K⁺-ATPase in the basolateral membrane (Giebisch, 1998; Köckerling et al., 1998).

K⁺ secretion (Fig. 5 B) is mediated by principal cells in the distal tubulus and cortical collecting duct (Wang, 1995; Köckerling et al., 1998). Na⁺ influx through amiloride-sensitive epithelial Na⁺ channels (ENac) in the apical membrane is rate limiting for K⁺ excretion. The driving force for Na⁺ influx is generated by the action of basolateral Na⁺/K⁺-ATPase which also provides the necessary K⁺ gradient. Both Na⁺ reabsorption and K⁺ secretion are stimulated by aldosterone. This hormone was shown to increase the apical Na⁺ conductance within hours and to upregulate expression of ENac and Na⁺/K⁺-ATPase (O'Neil, 1990). Intercalated cells (not shown), which constitute the second type of epithelial cells in these nephron segments, partially reabsorb urinary K⁺ via H⁺/K⁺-ATPase (Graber and Pastoriza-Munoz, 1993; Giebisch, 1998).

K⁺ secretion in the distal tubules is mediated by intermediate conductance (35 pS) weak inward-rectifier K⁺ channels (Wang et al., 1990, 1992). Their activity has been shown to be linked to basolateral Na⁺/K⁺ATPase by a pathway involving ion exchange, Ca²⁺ and PKC (Wang et al., 1993). In addition to regulation by protein kinases (Wang and Giebisch, 1991b) they have been shown to be particularly sensitive to changes in intracellular pH (pH_i) (Wang et al., 1990; Wang and Giebisch, 1991a). Intracellular acidification in the physiological range reversibly reduced channel open probability (Wang et al., 1990). An inward-rectifier K⁺ channel (31 pS) with a similar sensitivity to pH_i has been characterized in luminal membranes of the thick ascending limb of Henle's loop (Bleich et al., 1990).

These functional properties together with immunocytochemical data (Xu et al., 1997) confirmed, that $K_{ir}1.1$ channels underly the K^+ conductances involved in renal salt reabsorption and K^+ secretion.

1.2.4 The antenatal Bartter Syndrome (aBS)

Hereditary tubular disorders leading to excessive salt wasting are rare diseases with an incidence of approximately 1:50000 newborns. The first patients have been described by Bartter (Bartter et al., 1962) with symptoms of renal salt loss, hyperreninaemia, hyperaldo-steronism and hypokalemic alkalosis. In the following years, cases showing variants of this syndrome were reported by several groups (Gitelman et al., 1966; Fanconi et al., 1971; Seyberth et al., 1985). Currently, three types with biochemical and physiological characteristics that are similar to those resulting from long-term application of certain diuretics can be distinguished (Seyberth et al., 1997; Köckerling et al., 1998):

- antenatal Bartter Syndrome (aBS), more correctly termed Hyper-Prostaglandin E-Syndrome (HPS; (Konrad et al., 1999)) or FSLT (furosemide-like salt-losing tubulopathy)
- Gitelman-Syndrome; synonyms: Hypocalciuric Bartter Syndrome or TSLT (thiazide-like salt-losing tubulopathy)
- Pseudohypoaldosteronism Type 1 or ASLT (amiloride-like salt-losing tubulopathy)

This classification was confirmed by identification of the genes underlying these defects. Mutations in the gene coding for the thiazide-sensitive Na⁺/Cl⁻-cotransporter were found in patients with TSLT (Simon et al., 1996a). ASLT can be caused by mutations in the ENac channel subunits (Chang et al., 1996). The most severe form, aBS, is genetically heterogeneous: Mutations of either the furosemide-sensitive Na⁺/K⁺/2Cl⁻-cotransporter or *romk* have been identified in aBS patients (Simon et al., 1996b, c).

aBS is inherited as an autosomal recessive disorder. Mutations of *romk* appear to be rare, since the gene shows little allelic variation in the healthy population and most identified mutations are unique for single patients. Typically, affected children are compound heterozygous, i.e. they harbour a different mutation on either *romk* gene. Point mutations linked to aBS are apparently distributed over the whole coding region of the gene.

The molecular mechanisms leading to impairment or loss of $K_{ir}1.1$ channel function are yet unknown. A major pathophysiological consequence seems to be the impairment of NaCl reabsorption in the ascending loop of Henle which depends on K^+ recycling through $K_{ir}1.1$ channels ((Hebert, 1998), see also Fig. 5 A). The resulting loss of salt and fluid leads to polyhydramnios and premature birth (Seyberth et al., 1997). Polyuria remains life-threatening in young children if they are not treated by substitution with fluid, NaCl and KCl. PGE_2 is markedly elevated and plays an important role in the pathogenesis as it aggravates salt and fluid loss. Renin and aldosterone levels are also increased, but are insufficient to counteract the loss of NaCl, as Na⁺ reabsorption by principal cells in the distal tubulus depends on the presence of functional $K_{ir}1.1$ channels in the apical membrane (Fig. 5 B). Other secondary symptoms are hypercalciuria, which may lead to nephrocalcinosis, and alkalosis possibly resulting from excessive H^+/K^+ exchange in the collecting duct to reduce K^+ loss (Seyberth et al., 1985, 1997; Köckerling et al., 1998).

1.3 Goal of this study

The main goal of this work was to resolve the molecular mechanism underlying pH-dependent gating of $K_{ir}1.1$. It included

- · characterization of conformational changes associated with gating,
- investigation of interactions of pH-gating with other factors,
- identification of molecular determinants involved in pH_i sensing,
- · investigation of stoichiometry and cooperativity of pH-gating and
- verification of general implications for structure and function of other K_{ir} proteins.

As a result, a structural model of the pH-dependent gating mechanism was established and tested with respect to its applicability to the pathophysiology of aBS.

2. Results

2.1 pH-gating of K_{ir}1.1 is associated with conformational rearrangements

2.1.1 K_{ir}1.1 channels are gated by intracellular pH

Sensitivity to intracellular acidification has been demonstrated for both K^+ secretion channels in the kidney and cloned $K_{ir}1.1$ (ROMK) channels (Ohno-Shosaku et al., 1990; Wang et al., 1990; Tsai et al., 1995). The pH-dependent gating of $K_{ir}1.1$ channels was more closely investigated in giant inside-out patches from *Xenopus* oocytes. Solutions containing 120 mM K^+ buffered with HEPES (K_{int}) at pH values indicated were applied to the cytoplasmic side of the membrane patches. During the experiment every 800 ms the membrane potential of -80 mV was intermittently stepped to +50 mV for 50 ms. These experimental conditions were used for all patch experiments unless stated otherwise.

Intracellular acidification (from pH_i 8.0 to 6.0) led to complete channel closure within a few seconds, while alkalinization resulted in channel reactivation (Fig. 6). Under steady-state conditions, pH-gating showed half-maximal inhibition (pK_{app}) at pH_i 6.8 with a Hill coefficient of around 3 (Figs. 19 and 20).

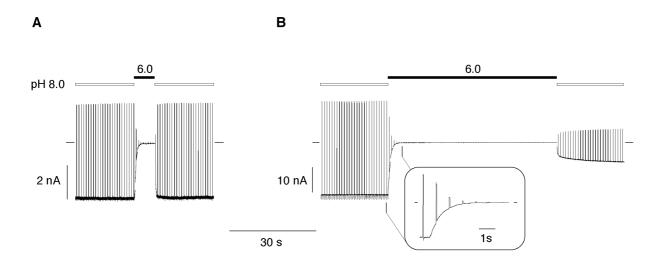


Fig. 6. (A), (B) pH_i-induced inactivation of K_{ir}1.1 currents in excised inside-out patches. Voltage step protocol and pipette solution as described in the text, application of solutions as indicated by bars (filled white: pH 8.0; black: pH 6.0). Inset: kinetics of pH-induced inactivation at an enlarged time scale. Note that the current amplitude at pH 8.0 remained constant, i.e. there was no "channel rundown".

The effect of intracellular pH on $K_{ir}1.1$ single channel behaviour was studied in small inside-out-patches. In the recording in Fig. 7 currents through single $K_{ir}1.1$ channels are resolved.

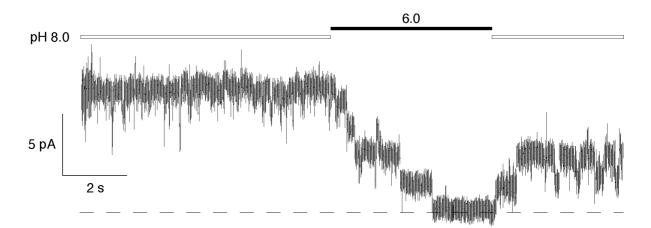


Fig. 7. Dependence of $K_{ir}1.1$ single channel amplitude on pH_{ir} . Current recording at high gain from an excised inside-out patch containing 4 active channels at a holding potential of -80 mV. The dashed line marks the current baseline (closing of the last active channel). Data were sampled at 10 kHz and filtered with 0.3 kHz. The patch pipette was filled with $K_{pipette}$ and intracellular solutions (K_{int}) were applied as indicated by bars.

The single channel amplitude (32.2 \pm 2.1 pS, n=4) was not affected by acidification. Instead, the decrease of the macroscopic current amplitude at pH_i 6.0 was due to a reduced channel open probability. In further control experiments, pH-induced inactivation was found to be independent of transmembrane voltage. The results were the same when a voltage protocol was used with a membrane potential of +50 mV intermittently stepped to -80 mV for 50 ms every 800 ms (data not shown). When the extracellular pH was varied from 8.0 to 6.0 no effect on $K_{ir}1.1$ currents was observed (data not shown). In summary, $K_{ir}1.1$ channels are gated specifically by intracellular protons in the physiological range without involvement of soluble cofactors.

After short periods of acidification recovery from pH-inactivation was complete, i.e. all channels reopened (Fig. 6 A). With longer periods of inactivation by low pH only a fraction of channels could be recovered (Fig. 6 B). This time-dependent loss of channel activity after pH-induced inactivation was reminiscent of 'channel run-down', a phenomenon well-known for K_{ir} channels. Therefore, MgATP (Fakler et al., 1994a; McNicholas et al., 1994) and the anionic phospholipid PIP2 (Hilgemann and Ball, 1996; Huang et al., 1998) both reported to counteract 'run-down', were tested.

As shown in Fig. 8 A, B neither reagent was able to restore channel activity lost after pH-induced inactivation. However, addition of dithiothreitol (DTT, $100 \,\mu\text{M}$) or reduced glutathione (GSH, 5 mM) to the pH 8.0 solution resulted in complete recovery even after prolonged acidification (Fig. 8 C, D). This suggested that oxidation occurred during acidification, which subsequently prevented channel recovery from pH-induced inactivation.

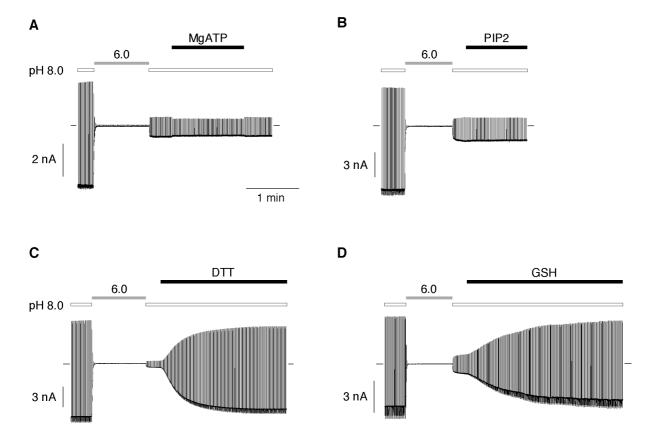


Fig. 8. Recovery of K_{ir}1.1 currents from pH-induced inactivation in excised inside-out patches. (**A**) with MgATP (e.g., free concentrations of 1 mM Mg²⁺ and 1 mM MgATP, see "Materials and Methods"); (**B**) with 0.1 mM PIP2; (**C**) 0.1 mM dithiothreitol (DTT); and (**D**) 5 mM glutathione (GSH). Voltage protocol as before, application of solutions and scalings as indicated.

2.1.2 Dependence of K_i, 1.1 channels on extracellular K⁺ is linked to pH-gating

The activity of $K_{ir}1.1$ channels does not only depend on pH_i but also on the concentration of extracellular K^+ , referred to as 'K⁺-regulation' (Doi et al., 1996), and both factors interact allosterically. In whole-cell experiments, outward currents in K^+ -free extracellular solution inactivated slowly and were completely restored in solution containing 90 mM K^+ (see also Fig. 12). The inactivation time constant increased by 50fold when the intracellular pH was lowered from pH_i 7.5 to 6.7. The activation time constant (switching to an extracellular solution containing 90 mM K^+) was independent from pH_i .

The interaction of extracellular K^+ with pH-gating of $K_{ir}1.1$ was subsequently investigated in giant inside-out patches. pH-induced inactivation of $K_{ir}1.1$ was completely reversible as long as solutions on either side of the membrane contained high concentrations of K^+ (120 mM) and DTT at the intracellular side (Fig. 8). As a variation from these conditions, K^+ in the pipette was replaced by Na^+ in the two experiments shown in Fig. 9.

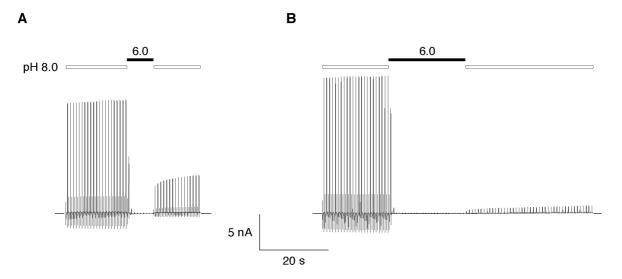


Fig. 9. (**A**), (**B**) Recovery of K_{ir}1.1 currents from pH-induced inactivation in the absence of extracellular K⁺ (excised inside-out patch). Pipettes were filled with K⁺-free solution (Na_{pipette}). Voltage protocol as before. The transient inward currents most likely result from local accumulation of K⁺ at the extracellular space after the depolarising pulse. All intracellular solutions contained 0.1 mM DTT.

Outward currents were stable at alkaline pH, but after pH-induced inactivation only a small fraction of channels recovered depending on the time spent in the closed state. This result could be explained in two ways: pH-gating changes the conformation of a K^+ binding site in the extracellular portion of the channel in a way, that K^+ is tightly bound in the open state but dissociates when the channel is closed. Alternatively, K^+ passing the pore could bind to an extracellular site stabilizing the channels' open state, and channel closure would prevent access of intracellular K^+ to this site. In the latter case $K_{ir}1.1$ channel activity would be expected to depend on the presence of intracellular K^+ when extracellular K^+ is absent.

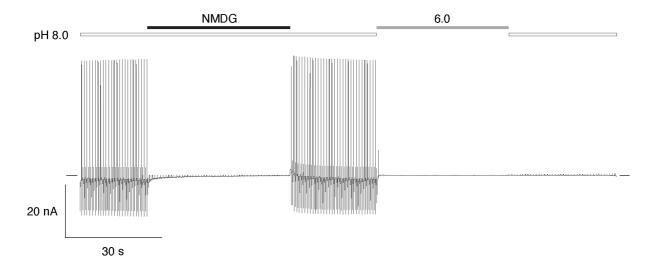
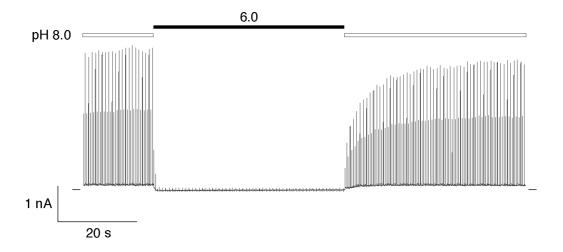


Fig. 10. Recovery of K_{ir}1.1 currents from complete removal of K⁺ at pH 8.0 (excised inside-out patch). Intracellular K⁺ was temporarily replaced by equimolar NMDG (non-permeant cations, 120 mM). Application of K_{int} pH 6.0 as control. All other experimental conditions as in Fig. 9.

The effect of complete K^+ deprivation was investigated in another experiment with K^+ -free pipette solution (Fig. 10). When intracellular K^+ was removed (application of 120 mM NMDG) no current could be observed. However, outward currents were completely and rapidly restored in K_{int} pH 8.0. Obviously, open channels (at pH 8.0) are either independent of K^+ or have a high affinity for K^+ , i.e. dissociation of K^+ is very slow (time constant > minutes). Subsequent application of K_{int} pH 6.0 for the same period of time led to an almost complete, irreversible loss of channel activity. Thus, dissociation of K^+ in the pH-inactivated state must be considerably faster (the time constant can be estimated to be in the range of seconds).

 $K_{ir}1.1$ currents could also be activated by extracellular Cs^+ which does not permeate the channel (Doi et al., 1996). This suggested that the interaction site is located extracellular to the selectivity filter. Indeed, dependence on extracellular K^+ was lost in a chimeric construct where the P-loop of $K_{ir}1.1$ had been replaced by a homologous sequence of K^+ -independent $K_{ir}2.1$ ($K_{ir}1.1$ -($K_{ir}2.1p$), Fig. 11). These channels were still gated by intracellular pH, i.e. currents inactivated completely at pH 6.0, but recovered upon alkalinization in the absence of extracellular K^+ .



However, when pH-gating was removed by an intracellular point mutation ($K_{ir}1.1(K80M)$), see section 2.2.3), dependence on extracellular K^+ was also abolished. This is demonstrated in whole-cell experiments which allow the exchange of extracellular solution. After one hour of preincubation in K^+ -deprived extracellular solution, $K_{ir}1.1$ currents had decreased close to background levels but slowly activated when switched to a solution containing 90 mM K^+ (Fig. 12 A). For the mutant $K_{ir}1.1(K80M)$ (Fig. 12 B), outward currents remained constant in K^+ -free solution and there was no additional increase in channel activity after switching to a high K^+ solution.

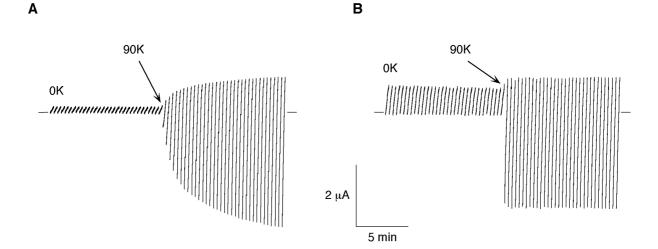


Fig. 12. Dependence of **(A)** K_{ir}1.1 currents and **(B)** K_{ir}1.1(K80M) currents on the presence of extracellular K⁺ recorded from whole oocytes. Two-electrode voltage-clamp protocol as described in "Materials and Methods". Note the outward currents **(B)** reflecting the weak inward-rectification of K_{ir}1.1 channels. 0K: K⁺-free bath solution, 90K: bath solution containing 90 mM K⁺.

Taken together, pH-induced inactivation is a prerequisite for the dissociation of extracellular K⁺ from the channel. This implies that pH-gating leads to conformational changes in the extracellular P-loop.

2.1.3 State-dependent cysteine modification reveals a conformational change induced by pH-gating

Sensitivity of K_{ir}1.1 channels to redox reactions was investigated with reagents that differentially react with sulfhydryl-groups. Oxidizing agents like Cu(II)-1,10-phenanthroline, which induce formation of disulfide bonds between cysteine residues (Kobashi, 1968), largely reduced the fraction of channels that spontaneously recovered upon realkalinization. Addition of DTT to the pH_i 8.0 solution still resulted in complete recovery from pH-induced inactivation (Schulte et al., 1998). However, when reagents were applied that can also oxidize single cysteine residues (in the following referred to as "modification") such as MTSES (Akabas, 1992) or DTNB (Riddles et al., 1979), DTT failed to recover channels from pH-induced inactivation (Fig. 13; experiment with MTSES not shown). The residual recovery of currents observed upon realkalinization in these experiments most likely represented unmodified channels since no recovery was seen when application of DTNB/pH 6.0 was extended to periods longer than 2 min (experiment not shown).

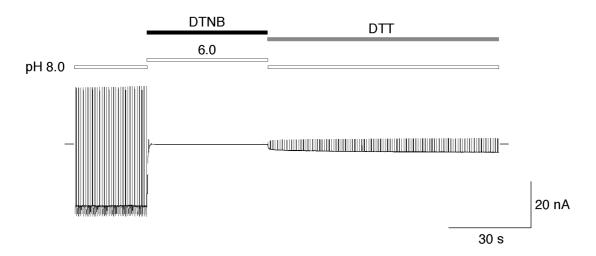


Fig. 13. Modification of K_{ir}1.1 channels with DTNB in excised inside-out patches. Application of 0.1 mM DTNB at pH 6.0 and 0.1 mM DTT at pH 8.0 as indicated by bars. Voltage pulse protocol as in Fig. 11.

These experiments demonstrate that formation of disulfide bonds in $K_{ir}1.1$ channels is reversible, while modification by DTNB or MTSES irreversibly locks pH-inactivated channels in a closed state. In order to more closely characterize modification by DTNB and its relation to pH-dependent gating, experiments were performed where DTNB was applied at pH $_i$ 8.0 either before or after pH-induced inactivation, i.e. when channels were either in the open or pH-inactivated (closed) state.

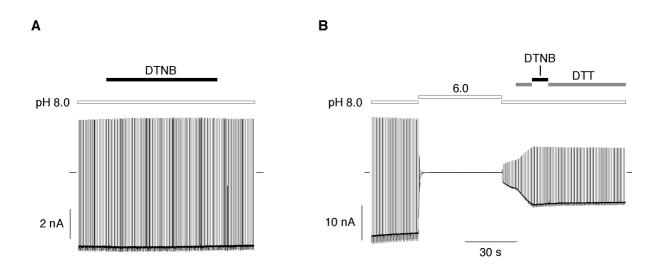


Fig. 14. DTNB-modification of open (**A**) and closed (**B**) K_{ir}1.1 channels at pH 8.0 in excised inside-out patches. Application of 0.1 mM DTNB at pH 8.0 and 0.1 mM DTT at pH 8.0 as indicated by the bars. Note that already a short time of DTNB-application prevented any further recovery of K_{ir}1.1 channels in 0.1 mM DTT (B).

As shown in Fig. 14, DTNB did not affect open channels at pH_i 8.0, nor did preapplication of DTNB change subsequent pH-gating (experiment not shown). In contrast, when DTNB was applied during DTT-induced recovery from pH-inactivation, recovery was promptly stopped

and did not continue in DTNB-free solution subsequently applied. Thus, under identical conditions, channels were only susceptible to chemical modification when they were pH-inactivated (i.e. in a closed state) prior to DTNB-application.

This coupling of chemical modification to a pH-inactivated state was confirmed by experiments with the mutant $K_{ir}1.1(K80M)$ in which pH-gating is abolished (see section 2.2.3). As shown in Fig. 15, no effect of DTNB was observed, neither at basic nor at acidic pH_i.

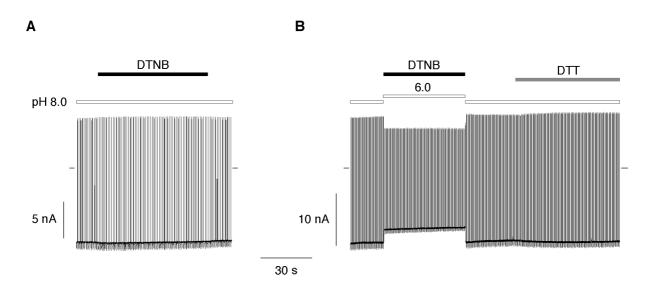


Fig. 15. Modification of K_{ir}1.1(K80M) channels with DTNB at (**A**) pH 8.0 and (**B**) pH 6.0 in excised inside-out patches. Application of 0.1 mM DTNB pH 8.0 or pH 6.0 and 0.1 mM DTT pH 8.0 as indicated by the bars.

Taken together, $K_{ir}1.1$ channels are targeted by DTNB in a state-dependent manner, i.e. channels are modified in the pH-inactivated closed state, but not in the conducting open state.

2.1.4 Cysteines 49 and 308 are targets for state-dependent modification

State-dependence together with the fact that DTNB specifically modifies cysteine residues was exploited to investigate which domains of the $K_{ir}1.1$ protein move during pH-dependent gating. For this purpose all cysteines in the $K_{ir}1.1$ sequence were replaced by alanine (A) or serine (S) (Fig. 16). Mutations in the N- and C-termini outside the 'core-region' (hydrophobic transmembrane domains and the P-region) resulted in functional channels gated by intracellular pH, while no currents were observed upon expression of the two mutants carrying C A/S exchanges in the P-region ($K_{ir}1.1(C121A,S)$ and $K_{ir}1.1(C153A,S)$).

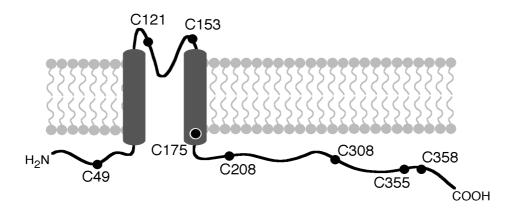


Fig. 16. Distribution of cysteine residues in the K_{ir}1.1 subunit. C121 and C153 are present in all K_{ir} subunits; C49, C208 and C308 are conserved in most K_{ir} subunits; C175, C355 and C358 are only found in K_{ir}1.1.

The redox properties of functional mutants were tested with a standardized protocol. Recovery from pH-inactivation in the presence of DTT was measured after a 50 s application of pH_i 6.0 solution either with or without (control) DTNB. Relative changes of current amplitudes were calculated as described in "Materials and Methods" for a quantitative comparison.

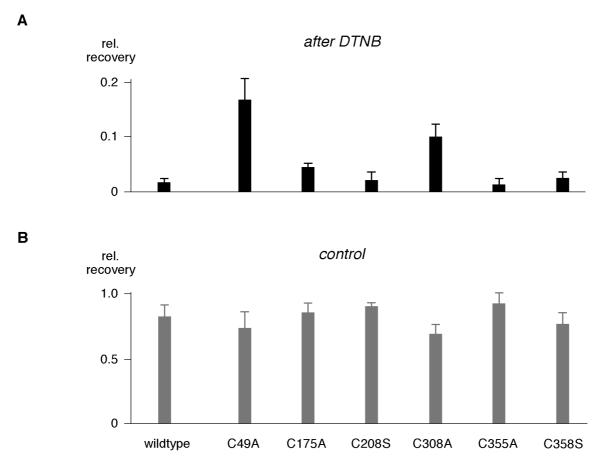


Fig. 17. Relative recovery of K_{ir}1.1 currents with DTT in excised inside-out patches after application of (**A**) pH 6.0 + 0.1 mM DTNB and (**B**) pH 6.0 alone (control) for 50 s each. Experimental conditions as in Fig. 13. There were no significant differences, no matter whether C A or C S mutants were tested. Calculations from n 3 experiments see "Materials and Methods".

All mutant channels recovered in a way very similar to that of $K_{ir}1.1$ wildtype channels under control conditions, and modification by DTNB was not abolished by any of the single C A/S exchanges (Fig. 17). However, the fractional recovery from inactivation after DTNB-modification observed in $K_{ir}1.1(C49A)$ and $K_{ir}1.1(C308A)$ was significantly larger than that for $K_{ir}1.1$ wildtype or any of the other mutants. A possible explanation was that DTNB-modification may occur at more than one site. To test for this, a double mutant $K_{ir}1.1(C49,308A)$ was constructed and investigated under the same conditions.

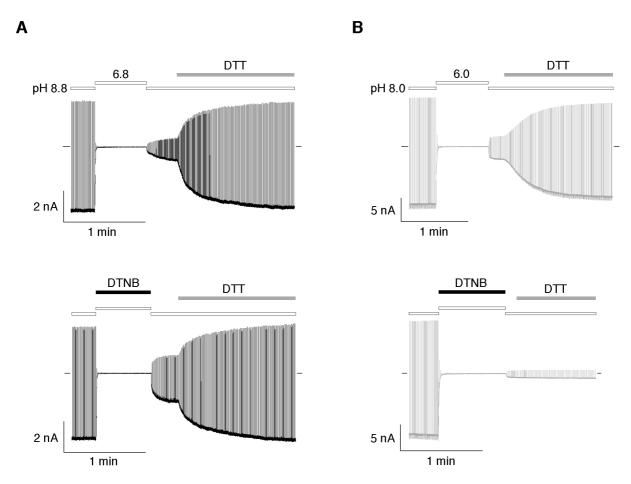


Fig. 18. (A) K_{ir}1.1(C49,308A) current recovery after application of pH 6.8 (upper panel) and pH 6.8 + 0.1 mM DTNB (lower panel) in excised inside-out patches. (B) For comparison, K_{ir}1.1 current recovery after application of pH 6.0 (upper panel) and pH 6.0 + DTNB (lower panel). All other experimental conditions as before.

Like in the case of $K_{ir}1.1$ and the single cysteine mutants only a fraction of channels recovered spontaneously from pH-induced inactivation (Fig. 18 A and B, upper panels). But in contrast to $K_{ir}1.1$ wildtype channels, recovery from pH-inactivation in the presence of DTT was complete for the double mutant, independent of whether DTNB had been added to the pH 6.0 solution or not (compare lower panels in Fig. 18 A and B).

These results show, that the cysteine residues modified by sulfhydryl-reagents in a state-dependent manner are C49 in the N-terminus and C308 in the C-terminus of the $K_{ir}1.1$ protein.

It may be concluded that pH-dependent gating in these channels is accompanied by structural rearrangements in both intracellular N- and C-termini. A noteworthy finding in this context was that $K_{ir}1.1(C49,308A)$ also displayed an alkaline shift in pK_{app} (see "Discussion").

2.2 Identification of lysine 80 (K80) as the sensor for pH_i in the neutral pH range

2.2.1 Current-pH_i-relation in K_{ir}1.1 and K_{ir}4.1 channels

For a quantitative analysis of pH sensitivity, concentration-responses of $K_{ir}1.1$ currents were measured in giant inside-out patches. Intracellular pH was varied between pH 8.0 (maximum current = reference pH) and pH 6.0 (no channel activity) with increments of 0.4 pH units.

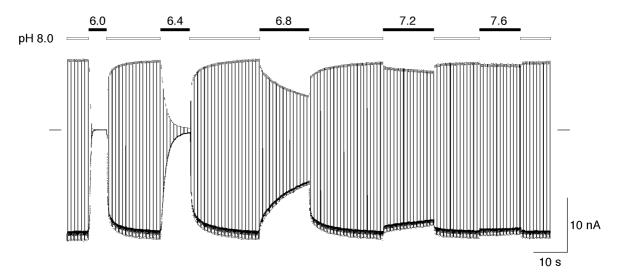


Fig. 19. pH titration of K_{ir}1.1 currents in an excised inside-out patch. All intracellular solutions contained 0.1 mM DTT and were applied at the pH_i values indicated by the bars. Note the pH-dependence of current inactivation kinetics.

The current amplitude was stable at alkaline pH and decreased reversibly upon acidification in a concentration-dependent manner (Fig. 19). The steady-state inhibition of the current was determined from monoexponential fits to the current decrease at a given pH_i and normalized with respect to the corresponding control current at pH_i 8.0. The resulting concentration-response curve showed half-maximal inhibition at pH_i (also referred to as pK_{app}) 6.8, with a Hill coefficient around 3 indicating cooperativity of the process (Fig. 20). $K_{ir}1.1a$ and $K_{ir}1.1b$ channels (data not shown) were indistinguishable with respect to pH-dependence and biophysical properties.

Negatively charged phospholipids have been shown to interact with several K_{ir} channels (Hilgemann and Ball, 1996; Huang et al., 1998) and to change gating properties of K_{ATP} channels (Baukrowitz et al., 1998; Shyng and Nichols, 1998). Therefore, $K_{ir}1.1$ pH concentration-response curves were measured before and after application of PIP2 to the cytoplasmic side of the patch membrane. As illustrated in Fig. 20, PIP2 had no significant effect on pH sensitivity of $K_{ir}1.1$ channels.

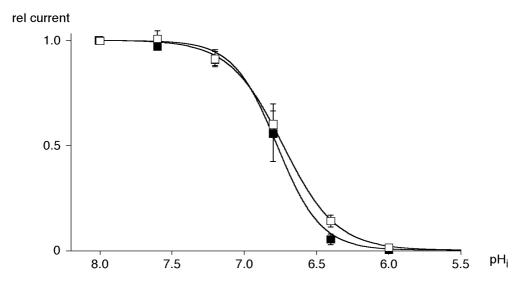


Fig. 20. pH_i concentration-response curves of $K_{ir}1.1$. Each data point represents the mean of steady-state activities in n experiments at the respective pH_i like in Fig. 19. Fits with a logistic function yielded pK_{app}=6.8 and Hill coefficient=2.9 (n=11) before (closed squares), and pK_{app}=6.7 and Hill coefficient=2.4 (n=6) after (open squares) application of PIP2 (10 μ M, 30 s). For details of data evaluation see "Materials and Methods".

Besides $K_{ir}1.1$, sensitivity to intracellular pH has also been reported for a strong inward-rectifier expressed in skeletal muscle (Blatz, 1984). $K_{ir}4.1$ (BIR10) is assumed to underly this conductance and was therefore tested in inside-out patches from *Xenopus* oocytes.

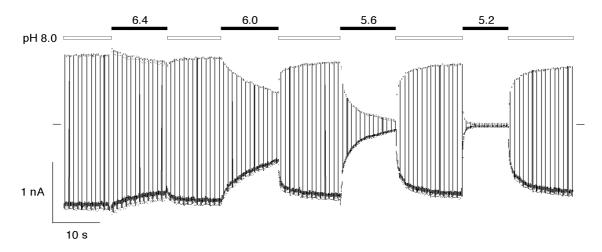


Fig. 21. pH titration of $K_{ir}4.1$ currents. Experimental conditions as in Fig. 19.

For $K_{ir}4.1$ the current-pH_i relation was of similar steepness as for $K_{ir}1.1$ (Hill coefficient 2.3), but shifted towards more acidic pH_i values (halfmaximal inhibition at pH 6.1, see Fig. 28).

2.2.2 pH sensitivity of other K_{ir} channels

Comparison of pH-gated versus pH-insensitive K_{ir} channels by primary sequence alignment should allow identification of molecular determinants required for pH-gating. Therefore, members of other K_{ir} subfamilies were tested for their sensitivity to pH_i. $K_{ir}2.1$ (IRK1), a representative of the subfamily $K_{ir}2$, did not show inhibition upon intracellular acidification in whole-cell experiments (Doi et al., 1996). Accordingly, $K_{ir}2.1$ channels remained active at low pH_i when examined in giant inside-out patches (Fig. 22).

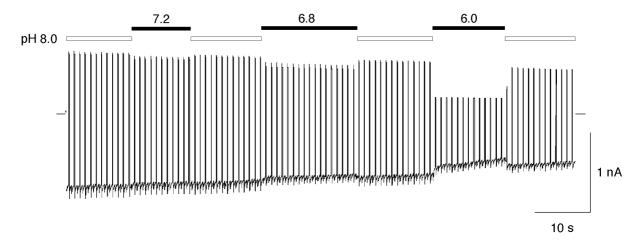
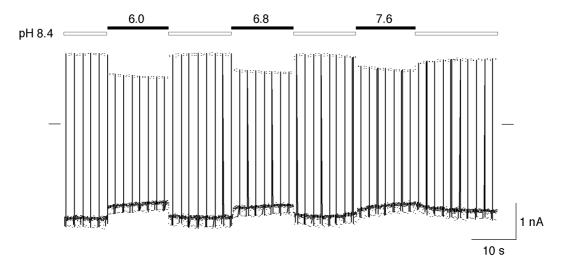


Fig. 22. pH sensitivity of $K_{ir}2.1$ currents in excised inside-out patches. Intracellular solutions (without DTT) were applied as indicated by the bars. Note that inward-rectification is lost in the absence of intracellular spermine and Mg^{2+} . The partial decrease of outward current at pH 6.0 was due to voltage-dependent block of the channel pore by intracellular protons.

Channels formed by $K_{ir}6.2 + SUR1$ are characterized by their high sensitivity towards intracellular ATP (therefore referred to as K_{ATP} channels) and serve to couple cellular excitability to metabolism (Ashcroft, 1988; Cameron and Baghdady, 1994). A possible modulatory role of intracellular pH in this process has been discussed (Misler et al., 1989; Bond et al., 1991; Proks et al., 1994). In whole *Xenopus* oocytes, changes in pH_i had complex effects on K_{ATP} currents (experiments not shown). Thus, pH_i-sensitivity of $K_{ir}6.2/SUR1$ currents was measured in excised inside-out patches in order to exclude any indirect influences of pH_i mediated by intracellular soluble factors. Prior to these experiments, 10 μ M PIP2 was applied to the patch for 20 s in order to prevent channel 'rundown'. Under these experimental conditions $K_{ir}6.2/SUR1$ currents were insensitive to intracellular acidification (Fig. 23).



<u>Fig. 23.</u> pH sensitivity of $K_{ir}6.2/SUR1$ currents after application of PIP2 (10 μ M, 20 s) in excised inside-out patches. Other experimental conditions as in Fig. 19.

Finally, members of the $K_{ir}3$ subfamily were tested in whole-cell experiments using the two-electrode-voltage-clamp technique (Fig. 24). pH_i was monitored by a third microelectrode filled with a proton-selective polymer (Tsai et al., 1995). Intracellular acidification was achieved through application of an extracellular bicarbonate solution (90 mM KHCO₃ pH 7.2; (Doi et al., 1996)). Uncharged H_2CO_3 and CO_2 molecules rapidly diffuse through the membrane and re-equilibrate with the cytoplasm.

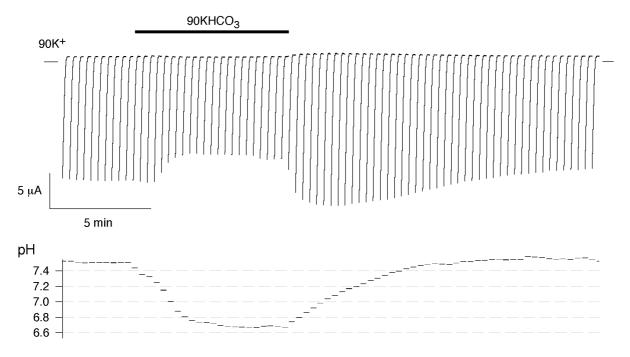
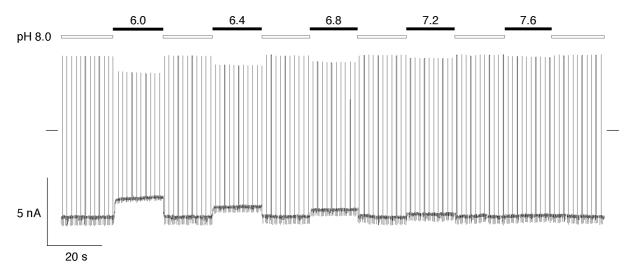


Fig. 24. Sensitivity of K_{ir}3.1/K_{ir}3.4 homomeric channels to intracellular acidification in whole *Xenopus* oocytes. Upper trace: currents in response to voltage-ramps (-120 to +50 mV, 20 s) showing strong inward-rectification. Lower trace: intracellular pH as recorded with a pH-sensitive microelectrode (see "Materials and Methods").

With switching to extracellular 90KHCO_3 , the resting pH_i (7.5) decreased to more acidic values (< 6.7) and reversed upon withdrawal of external HCO_3^- with a tendency to overshoot re-alkalinization to values around 7.6. As can be seen in Fig. 24, $K_{ir}3.1/K_{ir}3.4$ -mediated currents decreased only partially during acidification and recovered with a transient maximum upon re-alkalinization. $K_{ir}3.2$ (data not shown) exhibited a similar behavior in this type of experiments. Further investigations with excised inside-out patches could not be carried out since G-protein coupled K_{ir} channels quickly lost activity upon patch excision (fast "rundown", also reported by Ruppersberg et al. (1999)). From all K_{ir} channels tested, only $K_{ir}1.1$ and $K_{ir}4.1$ exhibited gating by intracellular protons.

2.2.3 Substitution of K80 by non-basic amino acids eliminates pH-gating of K_{ir}1.1 channels

In experiments with chimeric K_{ir} constructs, pH_i sensitivity could be conferred to $K_{ir}2.1$ by the N-terminus of $K_{ir}1.1$ (Fakler et al., 1996b). Consequently, N-terminal amino acid sequences of the K_{ir} channels tested above were compared and analyzed for titratable amino acid residues. By site-directed mutagenesis, lysine 80 (K80) was found to be a crucial determinant for pH-gating (Fakler et al., 1996b). $K_{ir}1.1$ channels where K80 had been replaced by a non-titratable amino acid ($K_{ir}1.1$ (K80T) or $K_{ir}1.1$ (K80M)) were no longer sensitive to pH_i (Fig. 25).



<u>Fig. 25.</u> pH titration of $K_{ir}1.1(K80M)$ currents in excised inside-out patches. Experimental conditions as in Fig. 19. Fits to the respective current-pH_i-relation with a logistic function yielded a pK_{app} of 5.3 and a Hill coefficient of 0.5 (n=5).

Thus, gating by pH_i and occurrence of a lysine residue at position 80 (or a homologous site) correlate well. All K_{ir} channels gated by pH_i exhibit a lysine residue at this site N-terminal to M1 (referred to as 'pre-M1 site'), all others show non-titratable amino acids at the homologous positions.

The results presented above led to the hypothesis, that protonation of K80 might trigger pH-gating. However, this would require that the pK_a of the NH_2 -group of this lysine residue is shifted by more than 3 pH units compared to its standard value of 10.5 (Dawson et al., 1986). To verify this, the protonation state of K80 was investigated by application of FmocCl, a reagent that reacts rapidly and specifically with $|NH_2$ -groups but not with protonated NH_3^+ -groups (Carpino and Han, 1970; Henczi and Weaver, 1994). Modification of unprotonated K80 would yield a urethane derivative which can no longer be protonated (Fig. 26 A).

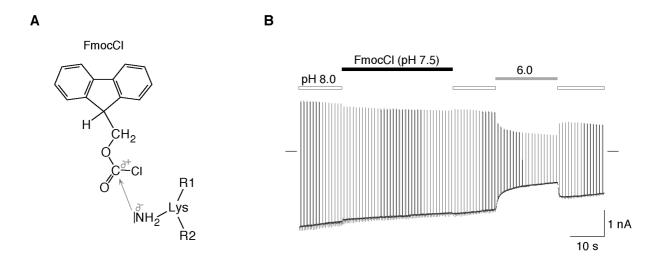


Fig. 26. (**A**) Reaction of FmocCl with NH₂ of lysine. Nucleophilic addition and subsequent elimination of Cl yields a urethane derivative. (**B**) Chemical modification of K_{ir}1.1 channels with FmocCl in an excised inside-out patch. FmocCl was freshly added to the modification buffer (MOD) from a stock solution (100 mM FmocCl in dioxane) for a final concentration of 0.1 mM. Solutions (K_{int} without DTT) were applied at the pH values indicated, other experimental conditions as before. Note that treatment of the patch with FmocCl did not alter channel activity.

As shown in Fig. 26 B, K_{ir}1.1 channels which had been treated with FmocCl at pH_i 7.5 prior to acidification were no longer inactivated at pH_i 6.0. The partial decrease in current observed upon acidification was most likely due to channels not modified by FmocCl.

From this experiment it can be concluded that K80 is not protonated at $pH_i > 7.5$, and that protonation of this residue is a prerequisite for pH-gating.

2.2.5 Introduction of lysine at the pre-M1 site confers pH-gating to other K_{ir} channels

Although K80 was necessary for pH-gating in $K_{ir}1.1$, it remained to be investigated if this determinant would be sufficient to induce pH-gating in other K_{ir} channels. Members of two K_{ir} subfamilies were mutated at the corresponding site $(K_{ir}6.2(T71K)/SUR1)$ and $K_{ir}2.1(M84K)$, and the resulting channels were tested for sensitivity to pH_i.

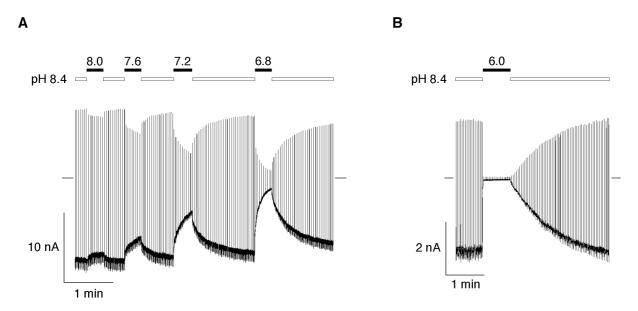


Fig. 27. pH sensitivity of (A) $K_{ir}6.2(T71K)/SUR1$ channels (after application of 10 μ M PIP2 for 20 s) and (B) $K_{ir}2.1(M84K)$ channels in excised inside-out patches. Experimental conditions as in Fig. 23. Note the time scales revealing slower kinetics of current recovery compared to Fig. 19.

As shown in Fig. 27, $K_{ir}6.2(T71K)/SUR1$ channels (A) displayed a high sensitivity to pH_i , as did $K_{ir}2.1(M84K)$ channels (B) (complete inhibition of channel activity at pH_i 6.0). Compared to $K_{ir}1.1$, pH-gating had different kinetics and was less reversible at low pH_i .

The concentration-response curves from pH titration experiments with $K_{ir}2.1(M84K)$ and $K_{ir}6.2(T71K)/SUR1$ are plotted in Fig. 28 together with those obtained for $K_{ir}1.1$ and $K_{ir}4.1$. All current-pH_i-relations displayed positively cooperative pH-gating although there was some variability in pK_{app} and Hill coefficient (see legend Fig. 28). The finding that pH-gating could be conferred to members of different K_{ir} subfamilies via introduction of a lysine at the pre-M1 site allowed several conclusions:

- Lysine at this site is sufficient (in the sense of a subunit-specific determinant) for pH-gating.
- A gating machinery is present in all K_{ir} channels, which is coupled to the charge of the amino acid side chain at this site.
- A chemical environment is structurally conserved among K_{ir} subfamilies that leads to strong shifts in the apparent pK of lysine at this site (pK_{app} ranges from 3.1 to 4.4 pH units).

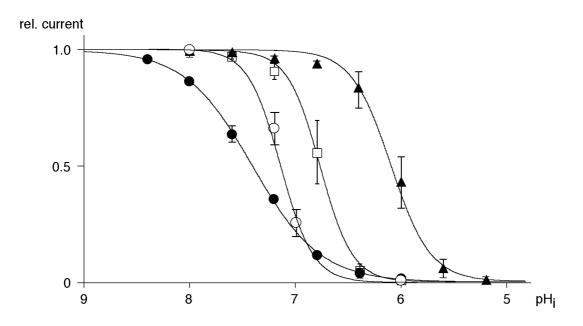


Fig. 28. pH concentration-response curves for pH-gated K_{ir} channels. Data points were obtained from pH titration experiments as in Fig. 19 as means of n steady-state values. Parameters of a logistic function fitted to the data were: pK_{app}=6.8 and Hill coefficient= 2.9 (n=11; taken from Fig. 20) for K_{ir}1.1 (squares), pK_{app}=6.1 and Hill coefficient=2.3 (n=6) for K_{ir}4.1 (triangles), pK_{app}=7.4 and Hill coefficient=1.4 (n=4) for K_{ir}6.2(T71K)/SUR1 (closed circles) and pK_{app}=7.1 and Hill coefficient=3.6 (n=5) for K_{ir}2.1(M84K) (open circles).

2.3 Intrasubunit assembly of K80 with R41 and R311 causes the shift in pK_{app}

2.3.1 Substitution of some conserved positively charged amino acids leads to shifts in pK_{app}

In order to work as a sensor for pH_i , lysine 80 must bind a cytoplasmic proton. Accessibility of this site from the cytoplasm has been confirmed by cysteine modification experiments for $K_{ir}2.1(M84C)$ (Fakler et al., 1996b). This evidence, together with the finding that the apparent affinity for H^+ is more than 1000fold lower than the standard value, led to the assumption that in the context of the $K_{ir}1.1$ protein K80 might be shielded by a positively charged environment repelling protons from its NH_2 -group. Accordingly, neutralization of such an environment should shift titration of K80 back to its standard value and thus result in channels permanently inactivated in the neutral pH-range (depicted in Fig. 29 A). To test this hypothesis, all intracellular arginine and lysine residues conserved among the K_{ir} subfamilies analyzed in 2.2.2 were replaced by glutamine (Q). The gating parameters of the mutant channels were determined from pH concentration-response measurements.

Α

$$X-\mathbf{R/K}+\begin{array}{c} H^{+} & R2 \\ H & CO \\ IN-(CH_{2})_{4}-C-H \\ H & NH \\ H^{+} & R1 \end{array} \longrightarrow \begin{array}{c} X-\mathbf{Q} & H-\overset{\mathbf{R}2}{N}-(CH_{2})_{4}-C-H \\ IN-(CH_{2})_{4}-C-H \\ I$$

В

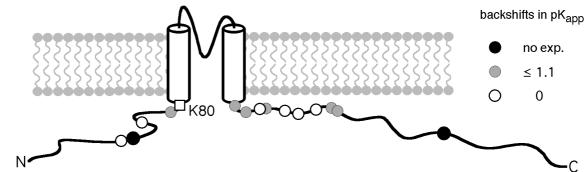


Fig. 29. (A) Electrostatic shielding of K80. X= backbone structure; grey zone: electrical field of the positive charge of R or K. When R or K are replaced by a neutral amino acid (Q), protons are no longer repelled from the NH₂-group of lysine (right).

(B) Neutralization of conserved intracellular arginine and lysine residues in K_{ir}1.1 (overview). Note that most mutations had no or only minor effects on pK_{app}.

As listed in Fig. 29 B and Table 1, most of the K/R $\,$ Q mutations resulted in channels with pH-gating either identical to $K_{ir}1.1$ wildtype or with a minor backshift in the respective p K_{app} . The latter mutations were located in regions close to the membrane, either N-terminal to M1 or C-terminal to M2.

<u>Table 1.</u> pK_{app} values of $K_{ir}1.1$ mutants. Concentration-response curves were obtained from at least 4 pH titration experiments and evaluated as described in Fig. 28. For $K_{ir}1.1(R41Q)$ and $K_{ir}1.1(R311Q)$ no channel activity could be detected.

channel mutant	рК _{арр}	channel mutant	pK _{app}	channel mutant	рК _{арр}
wildtype	6.8	K181→Q	7.5	K206 →Q	6.8
R39→Q	6.9	R184→Q	7.1	R212 → Q	6.7
R41→Q	-	K187→Q	6.7	R217 → Q	7.2
R61→Q	6.8	R188→Q	7.4	K218 → Q	7.1
R78→Q	7.9	R203→Q	6.8	R311→Q	-

2.3.2 K_{ir}1.1(R41Q) and K_{ir}1.1(R311Q) form permanently pH-inactivated channels

Neutralization of either positive charge at position 41 (R41Q) in the N-terminus or at position 311 in the C-terminus (R311Q) led to a complete loss of channel activity up to a pH_i of 10.0 (Fig. 30). The small currents at extremely high pH values most likely resulted from endogenous *Xenopus* ion channels since they were also detectable in patches from control oocytes (data not shown).

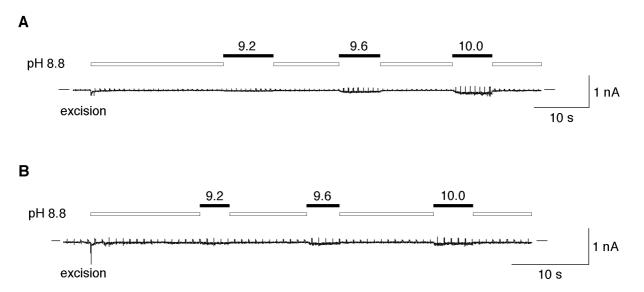


Fig. 30. Currents in excised inside-out patches from oocytes injected with **(A)** $K_{ir}1.1(R41Q)$ cRNA or **(B)** $K_{ir}1.1(R311Q)$ cRNA. Application of solutions ($K_{int}+0.1$ mM DTT) at alkaline pH as indicated by bars.

In order to verify that these mutants were actually translated and incorporated into the plasma membrane, they were tagged with a *flag* epitope (Wang et al., 1994) at their C-terminus and expressed in oocytes.

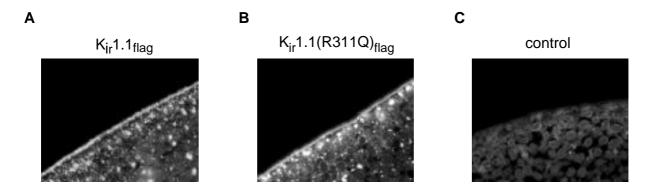


Fig. 31. Immunofluorescent staining of Xenopus oocytes injected with (A) $K_{ir}1.1_{flag}$ cRNA or (B) $K_{ir}1.1(R311Q)_{flag}$ cRNA; (C) a non-injected oocyte served as a negative control. Background fluorescense was negligible, whereas (A) and (B) showed strong fluorescense of the cell membrane and some intracellular vesicular structures. Slices (4 μ m) of fixed oocytes stained with anti-flag Cy3-labelled antibody were prepared as described in "Materials and Methods"; magnification approximately 200fold.

Immunostaining of slices from fixed oocytes with an anti-*flag* Cy3-labelled antibody showed that both mutants $(K_{ir}1.1(R311Q)_{flag};$ data for $K_{ir}1.1(R41Q)_{flag}$ not shown) were primarily localized in the plasma membrane (Fig. 31). Non-injected and $K_{ir}1.1_{flag}$ -injected oocytes served as negative and positive controls, respectively.

If the loss of channel activity in these mutations was attributable to permanent pH-inactivation, an exchange of K80 in $K_{ir}1.1(R311Q)$ or $K_{ir}1.1(R41Q)$ by a non-titratable methionine would be expected to result in 'recovery' of channel function. Indeed, both $K_{ir}1.1(R41Q,K80M)$ and $K_{ir}1.1(K80M,R311Q)$ expressed functional channels which were insensitive to pH_{i} - very similar to $K_{ir}1.1(K80M)$ (Fig. 32; compare to Fig. 25).

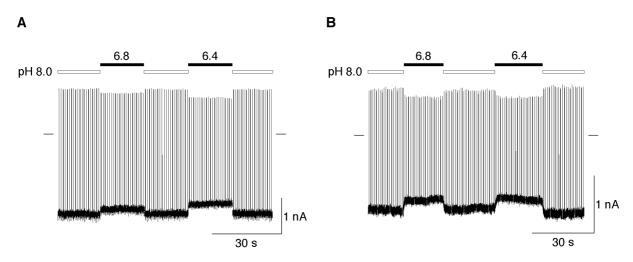


Fig. 32. pH sensitivity of **(A)** K_{ir}1.1(R41Q,K80M) and **(B)** K_{ir}1.1(R311Q,K80M) currents in excised inside-out patches. Experimental conditions like in the pH titration experiments before.

These results indicate that combined effects of arginines 41 and 311 are required to accomplish the anomalous titration of K80 in the $K_{ir}1.1$ protein.

2.3.3 Titration of K_{ir} 4.1(R294Q) shows a pK_{app} close to the standard pK of NH_2 -lysine

The experiments presented above suggested that neutralization of R41 and R311 resulted in closed channels due to permanent pH inactivation. However the possibility remained that these mutations affected the channel's gating machinery rather than shifting the pK_{app} of the pH_i sensor. Since this question could not be resolved with the $K_{ir}1.1(R41Q)$ and $K_{ir}1.1(R311Q)$ mutants, a homologous mutation was introduced in $K_{ir}4.1$ ($K_{ir}4.1(R294Q)$) and tested for channel activity at very alkaline pH.

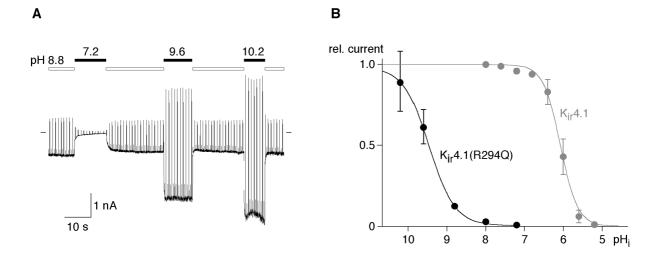


Fig. 33. (**A**) pH titration of $K_{ir}4.1(R294Q)$ currents in excised inside-out patches at alkaline pH. Experimental conditions as in Fig. 19, scalings as indicated. (**B**) pH concentration-response curve for $K_{ir}4.1(R294Q)$ as calculated from 5 experiments as in Fig. 33 A. Fit with a logistic function yielded pK_{app}=9.6 (Hill coefficient=2.1) in comparison to pK_{app}=6.1 (Hill coefficient=2.3) of $K_{ir}4.1$ wildtype (grey curve, taken from Fig. 28).

As demonstrated in Fig. 33 A, K^+ currents activated in excised inside-out patches and channels were still able to open and close at alkaline pH_i . Moreover, the resulting concentration-response curve (Fig. 33 B) showed that pH-gating was still positively cooperative, although it exhibited a pK_{app} (9.6) close to the standard value for lysine. These results indicate that neutralization of R41 or R311 specifically abolished the pK_{app} -shift of K80 but left the gating machinery intact.

2.3.4 Evidence for electrostatic interaction of R41 and R311 with the sensor for pH

Changing arginines to glutamines could result in structural or electrostatic changes in the chemical environment of K80. Conservative substitutions by lysine ($K_{ir}1.1(R41K)$, $K_{ir}1.1(R311K)$ and $K_{ir}1.1(R294K)$) were tested to elucidate the importance of positive charges at these sites. Fig. 34 shows that in $K_{ir}1.1(R311K)$ channels pH-gating was shifted to more alkaline pH values by about 0.3 pH units, while in $K_{ir}4.1$ the same mutation shifted pH-gating by 2 pH units. For $K_{ir}1.1(R41K)$, no channel activity could be observed up to a pH_i of 10.0. Thus, a positive charge at either determinant is not sufficient to explain the shift in pK_{app} observed for the gating-triggering lysine. Rather, they strongly suggest that lysine 80 and the two arginine residues are located in close proximity (an arrangement referred to as RKR triad (Schulte et al., 1999)), given that the mutagenesis effects illustrated in Fig. 34 were virtually caused by an exchange of a guanidino group versus an aminomethylene group.

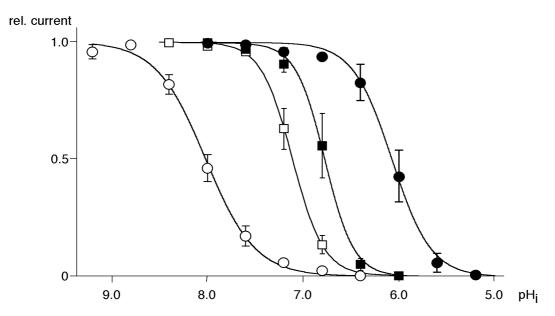


Fig. 34. pH concentration-response curves for K_{ir}1.1, K_{ir}1.1(R311K), K_{ir}4.1 and K_{ir}4.1(R294K) currents. pH titration experiments in excised inside-out patches were carried out as in Fig. 19. pH dose-responses were fitted with a logistic function to mean values of n experiments: pK_{app}=6.8 and Hill coefficient=2.9 for K_{ir}1.1 (n=11; closed squares; taken from Fig. 20), pK_{app}=7.1 and Hill coefficient=2.6 for K_{ir}1.1(R311K) (n=5; open squares), pK_{app}=6.1 and Hill coefficient=2.3 for K_{ir}4.1 (n=6; closed circles; taken from Fig. 28) and pK_{app}=8.0 and Hill coefficient=1.7 for K_{ir}4.1(R294K) (n=5; open circles).

2.3.5 The triad of K80, R41 and R311 is formed within an individual K_{ir} subunit

The pH_i-sensing triad of K80, R41 and R311 could either be formed within a single $K_{ir}1.1$ subunit or between N- and C-termini of different subunits. To resolve this question, the following coexpression strategy was used: cRNAs coding for a non-pH-gated (permanently open) subunit ($K_{ir}1.1(K80M)$) and a non-functional (permanently pH-inactivated) subunit ($K_{ir}1.1(R311Q)$ or $K_{ir}1.1(R41Q)$ - not shown) were co-injected. The latter was additionally 'labelled' with a determinant for strong voltage-dependent pore block by the polyamine spermine (N171D; (Fakler et al., 1994b)). This method had been successfully used for the quantification of heteromultimeric channel populations (Glowatzki et al., 1995).

Four possible configurations could result from heterotetramer formation, two of which are depicted in Fig. 35. An intrasubunit interaction of R41, K80 and R311 would prevent the formation of functional triads. Consequently, all channel populations would be expected to be pH-insensitive, independent of the number of positive charges at K80 required to close the pore. Alternatively, if N- and C-terminal domains of neighbouring subunits interacted, neutralization of R311 in $K_{ir}1.1(R311Q,N171D)$ would be compensated by R311 of $K_{ir}1.1(K80M)$ to form a functional triad. In this case, at least one population should form, that is pH-gated and can be blocked by spermine (SPM) in a strongly voltage-dependent manner. For a detailed discussion of strategy and evaluation of the results see section 3.5.1.

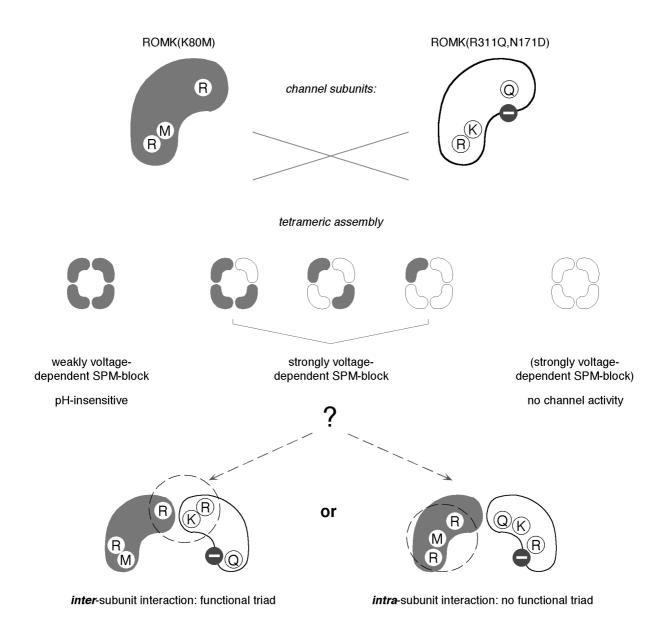
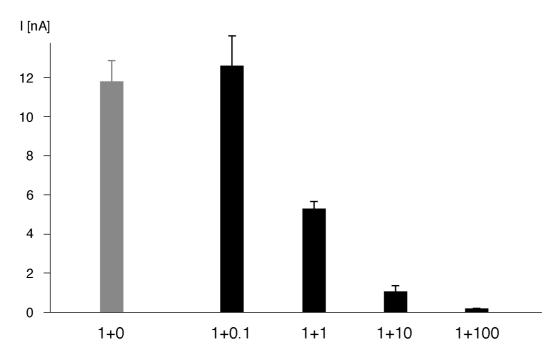


Fig. 35. Coexpression strategy to discriminate between *inter*- and *intra*subunit RKR triad formation. In white: K_{ir}1.1(R311Q,N171D) subunits - permanently closed; in dark: K_{ir}1.1(K80M) subunits - permanently open. All heteromeric channels carry at least one negative charge in M2 and are therefore excpected to be blocked by spermine with high voltage-dependence. Functional RKR triads only form upon intersubunit interactions. The sensitivity to pH_i of heteromeric channels could not be predicted.

Figs. 36 and 37 summarize the results of a series of coexpression experiments where a fixed amount of $K_{ir}1.1(K80M)$ cRNA had been injected together with variable quantities of $K_{ir}1.1(N171D,R311Q)$ cRNA in the ratios indicated. With increasing amounts of $K_{ir}1.1(N171D,R311Q)$ cRNA, average current amplitudes decreased from about 12 nA to less than 300 pA per patch. No suppression of $K_{ir}1.1(K80M)$ current was seen after co-injection of a control cRNA (coding for a non-functional channel) at a ratio of 1:10 (data not shown). Although the decrease did not reflect a binominal assembly, at least one non-functional heteromultimeric channel population must have formed. Similar results were obtained from coexpression experiments with $K_{ir}1.1(K80M)$ and $K_{ir}1.1(R41Q)$ (data not shown).



x equivalents of ROMK(K80M) cRNA + y equivalents of ROMK(R311Q,N171D) cRNA

Fig. 36. Quantitative analysis of current amplitudes (at -80 mV, pH_i=8.0) in excised inside-out patches resulting from coexpression of K_{ir}1.1(K80M) and K_{ir}1.1(R311Q,N171D) at the cRNA ratios indicated. Mean values of at least 3 oocytes (12 patches each) are presented.

The resulting currents were further analyzed for pH sensitivity and voltage-dependence of block by intracellular spermine.

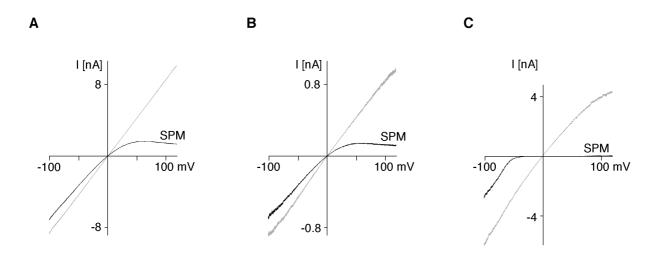


Fig. 37. Currents in response to voltage-ramps (-100 to +120 mV) measured in excised inside-out patches. In grey: K_{int} pH 8.0 - no inward-rectification; in black: K_{int} pH 8.0 + 1 mM SPM mediating voltage-dependent inward-rectification. (**A**) K_{ir}1.1(K80M) currents, (**B**) currents resulting from coexpression of K_{ir}1.1(K80M) and K_{ir}1.1(R311Q,N171D) cRNAs at a ratio of 1:10, (**C**) K_{ir}1.1(K80M,R311Q,N171D) currents as a positive control. Note the different current scales in (A) and (B).

Voltage-dependence of block was quantified by converting the current-voltage-recordings into conductance-voltage-plots and fitting them with Boltzmann functions as described in "Materials and Methods". Weakly voltage-dependent rectification mediated by 1 mM SPM could be seen with $K_{ir}1.1(K80M)$ in accordance with published results (Glowatzki et al., 1995). However, currents from all coinjection ratios displayed the same weak voltage dependence as $K_{ir}1.1(K80M)$ (shown for the example of the 1:10 cRNA ratio). The triple mutant $K_{ir}1.1(K80M,N171D,R311Q)$ harbouring a negatively charged aspartate in M2 was blocked with high voltage-dependence (onset of block was already visible a voltages negative from E_{K^+}) as expected and served as a positive control. Currents from all coexpressions displayed the same insensitivity to pH_i as $K_{ir}1.1(K80M)$ currents in the range of pH 6.8-9.6. This was tested in excised inside-out patches (n=12 each) as described in Fig. 33 A (data not shown).

These findings indicate that all currents resulted from $K_{ir}1.1(K80M)$ homotetramers and that all heteromultimeric channels formed were non-functional. This can only be explained by an intrasubunit assembly of the triad, as discussed in detail in section 3.1.5. In that case, any $K_{ir}1.1(N171D,R311Q)$ subunit would carry a permanent positive charge which could lead to channel closure. For the case of an intersubunit interaction, one or two functional triads would have formed, so the resulting channels should have been sensitive to acidic pH_i .

In summary, K80 together with R41 and R311 forms a triad within individual K_{ir} subunits. As a consequence of this structural arrangement protons are repelled from the NH_2 -group of K80 resulting in anomalous titration of this residue within the $K_{ir}1.1$ protein.

2.4 Defective pH-gating as a molecular mechanism for the pathogenesis of aBS

Disruption of the RKR triad by the R41Q and R311Q mutations in $K_{ir}1.1$ resulted in channel proteins that were incorporated into the plasma membrane, but were non-functional under physiological conditions (Fig. 30). Since such alterations in pH-gating might well explain impaired $K_{ir}1.1$ channel function under physiological conditions, the effect of known mutations associated with aBS on pH-gating was investigated. It is noteworthy that most of these mutations are clustered around the RKR triad or in domains close to it when considered in the context of the tertiary folding for K_{ir} proteins deduced from the results above (Fig. 38).

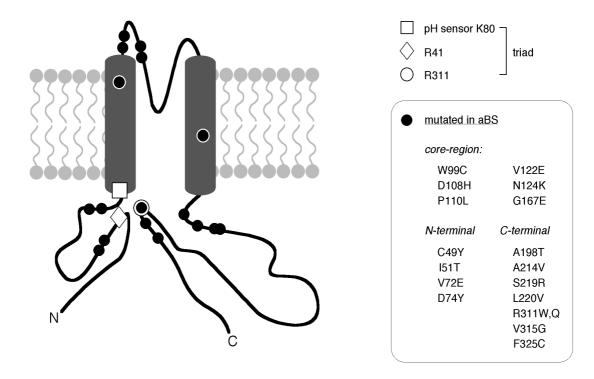


Fig. 38. Localization of known aBS mutations in a strucural model of K_{II}1.1. Determinants of the RKR triad (open symbols) are in close spatial proximity. Point mutations as taken from the literature: V72E, D74Y, W99C, D108H, P110L, V122E, G167E, A198T and V315G (Karolyi et al., 1997), N174K (Derst et al., 1998), A214V, S219R (Simon et al., 1996c); C49Y, I51T, L220V, R311W, R311Q and F325C (Schulte et al., 1999); see also "Materials and Methods" section 4.2.4.

Mutations in the intracellular N- and C-termini were expected to disturb the structural arrangement of the RKR triad, and in turn shift the pH-gating off the neutral range. This hypothesis was tested by heterologous expression of individual aBS mutant $K_{ir}1.1$ channels in *Xenopus* oocytes. Indeed, the majority of these mutations led to shifts in pH-gating towards more alkaline pH values (Fig. 39). The backshifts ranged from 0.3 pH units to > 2 pH units, with the largest shifts observed for mutations of the triad residues themselves (for example $K_{ir}1.1(R311W)$). $K_{ir}1.1(R311Q)$ did not express functional channels (Fig. 30). The ability of the channels to open and close was not affected in either of the aBS mutations investigated, nor was the cooperativity of the gating process altered significantly (Fig. 39).

It should be emphasized that a shift in pH-gating of as small as 0.3 pH units was able to induce severe clinical symptoms. For example, one patient (see "Materials and Methods" section 4.2.4) combined mutations L220F ($pH_{0.5}$: 0.3) with R311W ($pH_{0.5}$: 2.2) and another one harbored mutations I51T ($pH_{0.5}$: 0.5) and C49Y ($pH_{0.5}$: 0.7). In both cases, none of the parents (heterozygous carriers of a mutated allele) had shown symptoms of aBS.

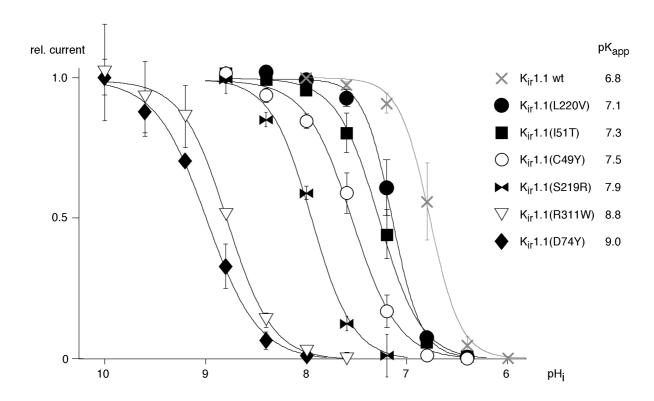


Fig. 39. pH concentration-response curves for currents of $K_{ir}1.1$ mutants identified in patients with aBS. pH titration experiments, data evaluation and fitting with a logistic function as described before (e.g., in Fig. 23). Note the wide range of backshifts in pK ($pK_{app} = 0.3$ to 2.2 pH units).

Taken together, these results suggest that structural disturbance of the RKR triad in $K_{ir}1.1$ leading to defective pH-gating is a major molecular cause for the pathogenesis of aBS.

3 Discussion

3.1 pH-gating of K_{ir} channels

3.1.1 Comparison of whole cell and patch experiments with physiological data

 $K_{ir}1.1$ channels display a high sensitivity to intracellular protons with acidification (pH_i 6.0) leading to complete channel closure. Since the single channel conductance was not affected by pH_i in the range of 6.0 - 8.0, the pH dose-response curves determined for macroscopic currents quantitatively reflected the dependence of 'channel open probability' on pH_i, referred to as pH-gating. Experiments with giant inside-out patches yielded titration curves with an apparent pK value of 6.8 and a Hill coefficient of 3.1 for $K_{ir}1.1$ (Fig. 20). The values for $K_{ir}1.1$ a were not significantly different from those for the splice variant $K_{ir}1.1$ b as confirmed by several groups (Fakler et al., 1996b; Choe et al., 1997; McNicholas et al., 1998). The pK_{app} values showed very little variation (0.1 pH units) when experiments were repeated with different batches of oocytes (5 series of titration experiments). The Hill coefficient ranged from 2.7 to 3.5 which may in part result from irreversible components of pH-induced channel inactivation. Therefore, only experiments showing more than 90% reversibility were considered for final evaluation. In these cases, the current inactivation time courses could be well fitted with a monoexponential function to determine steady-state inhibition (as described in section 4.5.4).

 pH_i -induced inactivation measured in whole oocytes was completely reversible in 90 mM K⁺ extracellular solution (Fakler et al., 1996b). More than 90% K_{ir}1.1 current inhibition was observed at $pH_i = 6.6$, which was monitored with a pH-sensitive microelectrode (as described in section 4.4.2). This value is higher than the 80% inhibition expected from the concentration-response curve at the same pH_i . It has to be considered, however, that the sensor for pH_i (lysine 80) is located at the intracellular surface of the plasma membrane and that the microelectrode can not sense the actual pH at that site. pH sensitivity of cloned $K_{ir}1.1$ has also been measured by others (Tsai et al., 1995; Choe et al., 1997; McNicholas et al., 1998) with very similar results. Thus, pH-gating is an intrinsic property of $K_{ir}1.1$ channels and is independent from intracellular factors.

K⁺ secretion channels in the apical membrane of tubular kidney cells have been shown to be particularly sensitive to intracellular acidification. In excised patches from rat principal cells (Wang et al., 1990), 'channel open probability' decreased from 0.96 at pH_i 7.4 to 0.15 at

 pH_i 7.0. In another study with inside-out patches from opossum kidney cells, K_{ir} currents displayed a pH_i -concentration-response with a pK_{app} of 7.5 (Ohno-Shosaku et al., 1990). These pK_{app} values are somewhat higher than those reported for cloned K_{ir} 1.1 channels, but the significance of these differences remains unclear.

So far, gating properties of $K_{ir}4.1$ channels have been poorly described in the literature. This study probably delivers a first quantitative description of gating by pH_i for this inward-rectifier K^+ channel. $K_{ir}4.1$ is expressed in a variety of tissues including skeletal muscle cells (Bond et al., 1994), for which a pH_i -sensitive inwardly rectifying K^+ current has been described ($pK_{app} = 6.1$, Hill coefficient = 3; (Blatz, 1984)). These results are in good agreement with the findings for cloned $K_{ir}4.1$ channels in the present study ($pK_{app} = 6.1$ and Hill coefficient = 2.3).

3.1.2 The role of K80 as a determinant for pH-gating

The results presented here establish that pH-gating of $K_{ir}1.1$ channels is determined by protonation of lysine 80. Replacement of this residue by a neutral amino acid (for example in $K_{ir}1.1(K80M)$) or $K_{ir}1.1(K80T)$) resulted in channels which were no longer gated by pH_{i} . Moreover, pH_{i} -sensitivity could be conferred on other members of the inward-rectifier family by changing the amino acid at the site homologous to lysine 80 (referred to as pre-M1 site). This was shown for $K_{ir}2.1$ and $K_{ir}6.2$ which are pH-insensitive. Of all K_{ir} subunits known to date, only members of the subfamilies 1 and 4 carry a lysine at this position. This correlates well with the pH sensitivity of these channels as described in sections 2.1.1 and 2.1.2. A minor pH_{i} -effect with negative cooperativity was found in some pH-insensitive K_{ir} channels like $K_{ir}2.1$ or $K_{ir}1.1(K80M)$. The nature of this effect most likely is a pore block by protons, which are present in micromolar concentrations at pH_{i} values < 6.0. These less pronounced effects of pH_{i} coincide with the presence of non-titratable amino acids at the homologous positions: asparagine (in $K_{ir}3.1$, $K_{ir}3.2$ and $K_{ir}3.4$), threonine (in $K_{ir}6$), arginine (in $K_{ir}3.3$) and methionine (in $K_{ir}2$, $K_{ir}5.1$ and $K_{ir}7.1$).

The role of lysine 80 as a pH_i sensor could be verified in a number of further experiments. Since protonation at this site induced channel closure (Figs. 7, 26), introduction of a "permanent" positive charge would be expected to result in closed channels. Indeed, when K80 in $K_{ir}1.1$ was replaced by arginine ($K_{ir}1.1(K80R)$), only minimal currents could be detected ($<2~\mu A$ at -100 mV and 90 mM extracellular K^+ in whole oocytes; data not shown). Correspondingly, $K_{ir}2.1(M84R)$ did not express functional channels, but $K_{ir}2.1(M84K)$ and $K_{ir}2.1(M84H)$ exhibited pH-gating in the neutral and acidic range, respectively (Fakler et al., 1996b). Replacement of this methionine 84 by any non-basic amino acid yielded functional $K_{ir}2.1$ channels which were insensitive to pH_i . Finally, the mutagenesis results were confirmed

by lysine modification experiments. FmocCl is known to selectively react with unprotonated amino groups and is therefore widely used in solid phase peptide synthesis as a protective group (Carpino and Han, 1970). -Amino-groups of lysine are effectively modified only if pH is kept above pH 9 (Henczi and Weaver, 1994). After FmocCl (0.1 mM) had been applied to excised inside-out patches for 30 s at pH 7.5, $K_{ir}1.1$ channels (which remained fully active under these conditions) were found to be insensitive against intracellular acidification (Fig. 26). The remaining partial reduction in current amplitude was probably due to incomplete modification which is not surprising as modification of lysine 80 in all four $K_{ir}1.1$ subunits is expected to be necessary to abolish pH-gating (see sections 3.1.4 and 3.1.5). In summary, protonation of lysine 80 is both required and sufficient to induce pH-inactivation in K_{ir} channels tested.

3.1.3 Mechanisms for shifts in pK_{app} as found for titratable residues within proteins

The finding that a lysine residue is titratable in the neutral pH range implies a shift in pK_{app} by more than 3 pH units (pKapp = -4.4 for $K_{ir}4.1$) compared to the standard value of 10.5 for NH_2 -lysine. In general, pK_{app} -shifts of titratable groups are thought to occur when amino acid side chains are placed in particular chemical environments within a protein, such as hydrophobic pockets, interact with other charges, or form hydrogen bonds. In this context, a decrease in pK_i by almost 4 pH units was assumed for either one or both of two lysine residues interacting via hydrogen bonds in ovotransferrin (Dewan et al., 1993), and an increase by about 3 pH units was reported for a histidine residue embedded in a highly positively charged environment within the FK506 binding protein (Yu and Fesik, 1994).

In K_{ir} proteins, the side chain of the amino acid at the pre-M1 site should be readily accessible from the cytoplasm, since lysine 80 in $K_{ir}1.1$ channels binds intracellular protons and cysteine 84 in $K_{ir}2.1$ (M84C) channels could be chemically modified by MTSEA (Akabas et al., 1992; Fakler et al., 1996b). Therefore, hydrophobic shielding of lysine 80 appears to be rather unlikely. Instead, the mutagenesis experiments performed in this study strongly suggest, that this lysine residue is embedded in a positively charged environment formed by arginines 41 and 311. These arginines are assumed to be permanently charged under physiological conditions, since their guanidino-groups have a standard pK_a value of 12.5 (Dawson et al., 1986). Conservative mutations (R K) introduced at positions 41 and 311 in $K_{ir}1.1$ and at position 294 in $K_{ir}4.1$ had very different effects. $K_{ir}1.1$ (R311K) channels displayed only a small backshift in pK_{app} (-0.3 pH units), but the homologous mutation in $K_{ir}4.1$ ($K_{ir}4.1$ (R294K)) had a more dramatic effect, as the apparent pK was shifted back by about 2 pH units. And for $K_{ir}1.1$ (R41K) no functional expression was found up to a pH_i of 10 (data not shown). This indicated that not only the charge of these side chains is critical for the protonation behaviour of

K80 but also their location relative to the NH_2 -group of lysine 80. Based on these results, it is concluded that K80, R41 and R311 are located in close proximity - an arrangement referred to as RKR triad (Schulte et al., 1999) - where the positively charged arginines establish a local field repelling protons from the lysine amino group. This field does not interfere with the transmembrane electrical field, since pH-gating was not voltage-dependent. Another position in the primary sequence was identified, from which charges exerted an electrostatic effect on pH-gating. Introduction of a lysine or glutamate at position 51 in $K_{ir}1.1b$ (equivalent to residue 70 in $K_{ir}1.1a$) shifted pH-gating by about 0.5 pH units to more acidic or alkaline pK_{app} values, respectively (Choe et al., 1997). Alternatively to electrostatic interactions, the arginine and lysine residues in the RKR triad might form hydrogen bonds similar to what has been reported for two lysines and a tyrosine residue in ovotransferrin (Dewan et al., 1993).

3.1.4 Positive cooperativity and stoichiometry of pH-gating

The mutagenesis results suggest a common molecular basis of pH-gating for all K_{ir} channels as further discussed in section 3.2. Key characteristics of pH-gating are a Hill coefficient > 2 indicating positive cooperativity and pH-dependent inactivation kinetics pointing to a complex gating reaction.

In general, positive cooperativity serves to increase the sensitivity of a stimulus-response coupled system within a narrow range of stimulus intensity. For K_{ir}1.1 the Hill coefficient was determined to be close to 3. Although this coefficient does not necessarily reflect positive cooperativity in the sense of conformational changes of proteins (Forsen and Linse, 1995), there are several lines of evidence that in the case of pH-gating of K_{ir} channels, the Hill coefficient indeed arises from cooperative interaction of the four subunits. Firstly, when pHgated K_{ir}1.1 and non-gated K_{ir}1.1(K80M) subunits were coexpressed, the measured pH_i doseresponse curves displayed a lower Hill coefficient (around 1.5) without a significant shift in pK_{app} (data not shown). Secondly, one of the $K_{\text{ir}}1.1$ mutants where a conserved lysine residue had been replaced by glutamine (K_{ir}1.1(K181Q); see Fig. 29) was pH-gated with a pK_{app} of 7.4 and a Hill coefficient close to 1. The inactivation kinetics of this mutant were very fast and independent of pH_i. It is tempting to speculate that in this mutant conformational transitions of individual subunit gates were uncoupled. This could explain the accelerated inactivation kinetics and may also account for the observed increase in pK_{app}. If this was indeed the case, it would imply that a single subunit gate in a closed conformation was sufficient to occlude the pore. Coexpression experiments with non-pH-gated K_{ir}1.1(K80M) and permanently pH-inactivated K_{ir}1.1(R311Q,N171D) subunits confirmed this "one gate" hypothesis. K_{ir}1.1(R311Q,N171D) cRNA had an inhibitory effect on the expression of functional K_{ir}1.1 channels (Fig. 36), which means that inactive heterotetramers must have formed. Secondly,

since no channel populations with a strongly voltage-dependent block by spermine were detectable, all heterotetramers formed must have been inactive (Fig. 36).

How do these results match with established concepts explaining cooperativity in multimeric proteins? According to the sequential model proposed by Koshland et al. (1966), all subunits would be in an active (open) conformation in the absence of ligands. Binding of H^+ induced an inactive (closed) conformation of the respective subunit and increased the H^+ -affinity of associated subunits. The "one gate hypothesis", however, is conflicting with the sequential model. Opening and closing (but not the protonation) of $K_{ir}1.1$ channels could be monitored electrophysiologically as an "all or none" transition - subconductance states were only rarely detectable in single channel experiments (compare the results of Choe et al. (1997) with those of Wang et al. (1990)). Since protonation of the first subunit can not be cooperative but would already close the channel, the resulting titration curve would be strongly distorted (not following a logistic function).

Alternatively, a model has been established in which only symmetrical transitions of subunit conformations are allowed to occur (Monod et al., 1965). The open conformation does not bind the ligand but participates in an equilibrium with a closed conformation in which the single subunits independently bind ligands. Such an "all or none" transition could explain some of the characteristics of pH-gating but this model also causes problems. With respect to the "one gate hypothesis" one could argue, that a permanently inactivated subunit couples to the conformations of other subunits in a way that forces the whole tetramer into a closed conformation. This would still be compatible with the finding that cooperativity is lost in K_{ir}1.1(K181Q), where conformations of subunits may not be strongly coupled any more. Moreover, this model would explain why assembly of permanently pH-inactivated and pHinsensitive channels (as tested for K_{ir}1.1(K80M) coexpressed with K_{ir}1.1(R311Q,N171D) or K_{ir}1.1(R41Q); see section 2.3.5) did not follow a binominal distribution. Assembly of subunits in closed and open conformations into K_{ir} channels would be energetically unfavourable. That would not be a problem in the sequential model. However, the symmetry model must also meet the kinetic aspects of pH-induced inactivation. If only subunits in the closed state bind intracellular protons, then the spontaneous transition open closed should become rate-limiting at low pH_i. In fast application experiments, which allow exchange of intracellular solutions at giant inside-out patches within a few milliseconds, the time constant of current inactivation did not saturate at low pH $_{\rm i}$ ($_{\rm inact}$ 580 ms at pH $_{\rm i}$ 6.0 and 246 ms at pH $_{\rm i}$ 5.0; B. Fakler, personal communication). A time constant lower than 200 ms for a symmetrical transition would be compatible with the finding that renal K⁺ secretion channels display one open state with a mean lifetime of 18.6 ms and one closed state with a mean lifetime of 0.7 ms at pH₂ 7.4 (Wang et al., 1990).

Apart from these considerations it should be kept in mind, that both theories on cooperativity had been developed for the case of multimeric proteins in which each subunit represents a functional unit. This is definitely not the case for K_{ir} channels, in which all *four* subunits are required for formation of *one* functional channel. Therefore, more detailed experiments combined with a different theoretical concept may be required to explain cooperativity of ion channel gating.

3.1.5 *Intra*- versus *inter*-subunit interaction

The identification of conserved arginines 41 and 311 closely interacting with lysine 80 in K_{ir}1.1 channels raised the question, how this arrangement is formed in the quarternary structure. Theoretically, there may be four possibilities to form an RKR triad, involving one, two or three subunits: (a) all residues within a single subunit interact, (b) K80 and R41 from subunit I interact with R311 from an adjacent subunit II, (c) K80 and R311 from subunit I interact with R41 from an adjacent subunit II, and (d) K80 from subunit I interacts with R41 from an adjacent subunit II and R311 from another adjacent subunit III. A strategy was designed to address this problem consisting of two series of coexpressions: non-pH-gated K_{ir}1.1(K80M) together with permanently pH-inactivated (i) K_{ir}1.1(R311Q) or (ii) K_{ir}1.1(R41Q). As already discussed in section 3.1.4, only currents indistinguishable from K_i,1.1(K80M) could be detected in these coexpression experiments indicating that all heterotetramers formed were nonfunctional in the pH_i range of 6.4 to 9.6. This means that at least one subunit in the heterotetramers must be permanently pH-inactivated. Only assembly (a) full-fills this condition for both coexpression series. In the case of (i), R311Q would be compen-sated by the corresponding R311 of K_{ir}1.1(K80M) yielding a functional triad (not permanently pHinactivated; see Fig. 35) according to the assembly models (c) or (d). Correspondingly, R41Q (ii) would be compensated by R41 of K_{ir}1.1(K80M) as a result of an assembly like in (b) or (d). Thus, K80, R41 and R311 form triads underlying pH-gating in the physiological range within individual K_{ir} subunits. This may have important consequences for possible compensation of mutations affecting pH-gating and for the understanding of subunit assembly.

Nevertheless it is expected that also N- and C-termini of *adjacent* subunits interact with each other. Several approaches have been taken to identify the sites responsible for subunit assembly into functional K_{ir} tetramers (Tinker et al., 1996; Woodward et al., 1997). In addition, open and closed subunit conformations must somehow interact to explain the positive cooperativity observed for pH-gating.

3.1.6 A model for pH-gating of K_{ir}1.1 channels

Altogether, at least one open and two closed conformational states of K_{ir}1.1 have been characterized in this study. In Mg²⁺-free alkaline (pH 8.0) solution K⁺ currents remained constant reflecting a stable open conformation of the K_{ir}1.1 protein. Protonation induced a conformational change that led to channel closure (Fig. 6). This reaction was very slow, and the rate of inactivation was strongly dependent on pH_i (see Fig. 19) as already discussed in section 3.1.4. But protonation-deprotonation reactions are the fastest diffusion-controlled reactions known in solutions, their rate constants ranging below 10 ns. Thus the kinetic behaviour could not be explained by a simple one-step reaction. It rather pointed towards a reaction sequence involving multiple conformational transitions. In fact, there must be at least two closed states as revealed by patch clamp experiments on the interaction of K⁺ regulation and pH_i-gating of K_{ir}1.1 (see section 2.1.2). When channels were inactivated by application of pH 6.0 for a few seconds, a significant fraction recovered instantaneously in the absence of extracellular K⁺ (Fig. 9 A). Thus a closed state must exist with K⁺ still bound to an extracellular part of the pore. Recovery from pH-induced inactivation failed with increased periods of inactivation. From such experiments, the time constant for K⁺ dissociation from pH-inactivated K_{ir}1.1 channels could be estimated to be somewhat less than 10 s. It seems unlikely, however, that dissociation of K⁺ is responsible for the slow inactivation kinetics, since pH-gating of K⁺independent K_{ir}1.1(K_{ir}2.1p) channels displayed very similar pH-dependent inactivation kinetics (with a pK_{app} of 6.6) as K_{ir}1.1 wildtype (data not shown). A preliminary reaction sequence is proposed on the basis of these results, which could deliver an alternative explanation for the positive cooperativity of pH-gating (compared with the sequential or the symmetric model discussed in section 3.1.4):

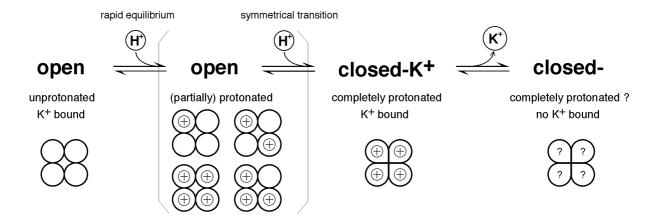


Fig. 40. Kinetic model of pH-gating of K_{ir}1.1 channels. In brackets: hypothetical channel states. Protonation is indicated by " ". Note that only the second transition (open closed) can be monitored directly. The protonation state of the K⁺-depleted closed state could not be determined.

The majority of open channels is unprotonated but participates in a rapid protonation-deprotonation equilibrium. Partially protonated channels would still be open for a short period of time before they are either deprotonated or undergo a symmetrical conformational change into a closed state. Positive charges at K80 in the open conformation are energetically unfavourable. Therefore, an increase in the number of positive charges would accelerate the transition into a closed state. The closed state must have a higher affinity for protons, since protonation of K80 is the source of energy for the gating reaction. Because of that, the closed state is believed to be highly protonated (indicated above by 4 positive charges). This would explain both the positive cooperativity and the pH-dependent inactivation kinetics of pH-gating. Evidently this reaction sequence combines aspects of both allosteric models discussed in chapter 3.1.4.

In its closed conformation, $K_{ir}1.1$ has a lower affinity for K^+ bound to an extracellular domain. As demonstrated in giant-inside-out patch experiments (see Figs. 9 and 10) this may be due to an increased off-rate for K^+ binding. The closed, K^+ -depleted channel can only be reopened if K^+ is present extracellularly and the cytosol is alkalinized subsequently. Structural aspects of these transitions are discussed further in section 3.2.1.

3.2 Structural implications for gating of K_{ir} channels

3.2.1 Conformational changes in K_{ir}1.1 induced by protonation

K_{ir} channels are presently conceived as ion-selective pores mainly controlled by intracellular blocking cations such as polyamines, Mg²⁺ or protons, but not undergoing significant conformational changes. In the present study two interdependent regulatory mechanisms are analyzed which imply conformational changes of the channel protein. pH-dependent opening and closing of K_{ir}1.1 channels is accompanied by conformational changes involving movement of the intracellular N- and C-termini. This was visualized by state-dependent modification of cysteines 49 and 308, which were susceptible to reaction with DTNB in the pH-inactivated state, while no modification was observed for open channels. It should be pointed out that these residues are close to the two conserved arginines that were found to underly the large pK-shift of K80. As concluded from a putative gating model (Fig. 40), K_{ir}1.1 channels should have a higher affinity for protons in the open than in the closed state. This could be achieved by movements of structural determinants of the RKR triad (as depicted in Fig. 41), which is in agreement with the finding that C49 and C308 move during gating. Conversely, mutations at these two sites and closeby (as found in aBS patients, for example; see section 3.3.3) display

slight shifts of pK_{app} to alkaline values, emphasizing the coupling of structural changes in these regions to the titration properties of K80.

Besides modification by DTNB or MTSES, channels were also oxidized in a state-dependent manner. Cu-Phen applied at pH $_{\rm i}$ 6.0 prevented that channels recovered from inactivation - an effect that could be fully reversed by application of the reducing agent DTT (Schulte et al., 1998). But although oxidation and reduction were induced by agents specific for formation and cleavage of disulfide bonds, no significant alteration of the redox-sensitivity was observed in either of the C A/S mutations. Moreover, oxidation was also observed in the double-mutant $K_{ir}1.1(C49,308A)$ which completely abolished state-dependent modification by DTNB (Fig. 18) as well as in the triple mutants $K_{ir}1.1(C308,355,358A)$ and $K_{ir}1.1(C49,175,308A)$ (data not shown). When all cysteines outside the 'core-region' were replaced by alanine, no channel activity could be detected. Thus, it was not possible to identify the cysteine residues involved in oxidation or to exclude a role of proteins closely associated with $K_{ir}1.1$.

 $K_{ir}1.1$ and $K_{ir}4.1$ channels are unique among inward-rectifiers in that they are dependent on extracellular ions like K^+ , NH_4^+ , Rb^+ or Cs^+ (but not Li^+ and Na^+), previously described as K^+ regulation (Doi et al., 1996; Pearson et al., 1999). The requirement of extracellular K^+ for channel activity was found to be dependent on the conformational state. At pH_i 8.0, $K_{ir}1.1$ channels showed no inactivation even in the complete absence of intra- and extracellular K^+ (see Fig. 10). In contrast, pH-inactivated channels (pH_i 6.0) could not be recovered in the absence of extracellular K^+ (as a function of time spent in the closed state; see Fig. 9). Thus, pH-induced inactivation either alters the affinity of an extracellular K^+ binding site or leads to a pore conformation that is then dependent on extracellular K^+ (not shown in Fig. 41). Further experiments will be needed to discriminate between these possibilities. In line with these findings, the non-pH-gated mutant $K_{ir}1.1$ (K80M) did not display any dependence on extracellular K^+ (Fig. 12). Taken together, these results provide a molecular explanation for the allosteric interaction of K^+ -regulation with pH_i -gating described for $K_{ir}1.1$ (Doi et al., 1996) as summarized in the structural model of Fig. 41.

So far, the extracellular binding site for K^+ could not be identified. The selectivity pattern for alkali ions raises the possibility, that the binding site may be located in a restrictive structure of the pore, possibly close to or in the selectivity filter. Consequently, replacing the pore-forming loop of $K_{ir}1.1$ by that of $K_{ir}2.1$ in a chimeric construct ($K_{ir}1.1(K_{ir}2.1p)$) abolished dependence on extracellular K^+ without affecting gating by pH_i significantly (Fig. 11). Furthermore it remains to be elucidated in which way the conformational changes resulting from pH-dependent gating may be linked to other processes reported to affect channel activity such as phosphorylation by protein kinase A (Xu et al., 1996) or interaction with anionic phospholipids (Huang et al., 1998).

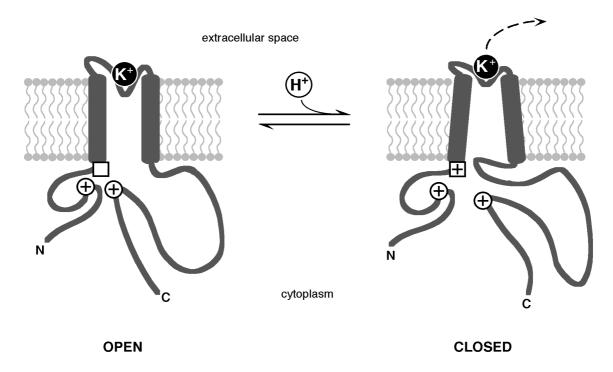


Fig. 41. Structural model of pH-gating of K_{ir}1.1 channels. Circles: arginines 41 and 311 (permanently positively charged); squares: lysine 80 (pre-M1 site). Conformational changes during pH-gating may rearrange the RKR triad and affect the interaction of extracellular K⁺ with the P-loop.

3.2.2 Comparison of pH-gating with gating mechanisms in other K_{ir} channels

pH-gating could be introduced into members of different K_{ir} subfamilies by mutating the pre-M1 site to a lysine residue (Fig. 27). Although the respective pK_{app} values displayed some variability, titration of the NH_2 -group in all these subtypes was shifted down by more than 3 pH units compared to its standard value. Thus, all K_{ir} proteins basically offered the structural context necessary for formation of the RKR triad and the associated gating machinery. Regarding the primary structure this may not be too surprising, since both arginine residues are conserved throughout the K_{ir} family and reside within the region where K_{ir} proteins show a very high degree of homology. Taking into account the sensitivity to minor side chain variations (Fig. 34), this result points to a significant conservation among K_{ir} subunits also on the level of their tertiary structure. Although deduced from functional analysis, these results suggest a first rough framework for the folding of K_{ir} proteins, in that N- and distal C-termini are backfolded to the pre-M1 region (Fig. 41). In line with such an interpretation, some studies reported involvement of both N- and C-termini in G protein-dependent gating of K_{ir} 3 channels (Huang et al., 1997) or inhibition of K_{ATP} channels by ATP (Tucker et al., 1998).

 K_{ir} channels may possess different trigger mechanisms linked to the same or structurally separate "gates". For example, $K_{ir}6.2(T71K)/SUR1$ channels are pH_i -gated (Fig. 28) and completely inhibited by ATP at intracellular concentrations of 1 mM (Ruppersberg et al., 1999).

Application of 10 μ M PIP2 (20 s) strongly reduced the sensitivity to intracellular ATP (data not shown) but did not abolish gating by pH_i. For K_{ir}1.1 is was demonstrated that protonation and 'inactivation-gating' are actually mediated by separate determinants. In K_{ir}1.1(R311W) and K_{ir}4.1(R294Q) mutant channels the working range of the 'pH-sensor' was shifted back closely towards the standard value of NH₂-lysine with the gating machinery left intact (Figs. 33 and 39). Thus, the RKR triad presented here explains the molecular mechanism underlying the 'driving-force' of pH-gating, but the localization of the gate remains open.

Recently, gating induced by changes in extracellular pH (pK_{app} around 2.5) has been related to conformational changes in the two segment-type K^+ channel from *Streptomyces lividans* (Perozo et al., 1998). Using EPR-spectroscopy, Perozo and coworkers uncovered structural rearrangements at the C-terminal end of the second transmembrane helix which were hypothesized to change the width of the inner vestibule and to thereby control ion permeation. Whether such a mechanism also underlies pH-gating in $K_{ir}1.1$ remains to be elucidated. Alternatively, the gate may be formed by protein domains neighbouring the RKR triad, since these domains including cysteines 49 and 308 were shown to move during pH-gating.

Activation of $K_{ir}3$ channels by G proteins as well as regulation of K_{ATP} channels by sulfonylureas and channel openers are mediated by accessory proteins interacting with the K_{ir} subunits. Consequently, these gating mechanisms can be classified as being *extrinsic* with respect to the channels. In contrast to that, pH-gating is *intrinsic* to the $K_{ir}1.1$ and $K_{ir}4.1$ proteins, i.e. it does not require any accessory -subunit or cofactors such as anionic phospholipids. As shown in Figs. 8 and 20, $K_{ir}1.1$ channels did not display 'run-down' in these experiments, nor did application of PIP2 change steady-state parameters of pH-gating significantly. Furthermore, pH-gating was also observed in $K_{ir}1.1(R188Q)$ mutant channels, which were reported to have reduced binding affinity for PIP2 (Huang et al., 1998). The mild shift in pK_{app} found for this mutant was similar to that seen in other N- and C-terminal mutations (see Fig. 29).

3.3 Physiological importance

3.3.1 Redox sensitivity of K_{ir}1.1 channels

The presence of reducing agents like DTT (0.1mM) and GSH (5 mM) was required for complete recovery from pH-induced inactivation in excised inside-out patches. In living cells, however, the redox-sensitivity most likely is not of physiological importance, since reduced glutathione should be available in millimolar concentrations. Nitric oxide has been reported to act as a physiological modifying agent of thiol groups. For example, NO could be replaced by

DTNB in activation of cyclic nucleotide-gated channels (Broillet and Firestein, 1996, 1997). However, in inside-out patch experiments nitric oxide donors (SIN-1 and S-nitrosocysteine, see "Materials and Methods") did not reveal any effect on pH-gating of K_{ir}1.1 channels (data not shown). Thus it remains unclear if state-dependent reactivity of these conserved cysteines has functional consequences *in vivo*.

3.3.2 K⁺ secretion in the kidney

Only 1-2 % of total body K⁺ is located in the extracellular fluid, the vast majority is contained by cells (Stanton and Giebisch, 1992). As a consequence, extracellular K⁺ levels can be strongly increased by dietary intake and internal redistribution. Under normal conditions, rapid changes of extracellular K⁺ levels are effectively buffered by cellular uptake and secretion of K⁺. Long-term K⁺ homeostasis is achieved by regulation of renal K⁺ secretion (Giebisch, 1998). A major source of disturbance is the acid-base status in the body, since pH and extracellular K⁺ levels are tightly coupled. During acidosis, for example, Na⁺/H⁺-antiport in cells is activated and the depletion of intracellular Na⁺ impairs Na⁺/K⁺-ATPase leading to extracellular accumulation of K⁺ (Lang and Rehwald, 1992; Giebisch, 1998). Disturbance of K⁺ blood plasma levels can be life-threatening, mainly because of evoked cardiac arrhythmias. Other common symptoms of hyperkalemia are muscular weakness, impaired peristaltic movements of the gut and metabolic alkalosis (Thier, 1986).

The interdependence of K^+ and pH requires a complex regulation of kidney function. Renal K^+ secretion is enhanced during metabolic alkalosis and inhibited during metabolic acidosis. The latter is an important mechanism to prevent excessive loss of K^+ during acidosis (Wang, 1995). pH-gating of $K_{ir}1.1$ channels represents a molecular mechanism linking both K^+ and pH:

- Since K_{ir}1.1 channels are responsible for K⁺ recycling into the lumen of the thick ascending loop of Henle (Fig. 5 A), their inhibition by acidic pH_i would lead to enhanced K⁺ reabsorption. The same effect would result from inhibition of K⁺ secretion in principal cells of the cortical collecting duct (Fig. 5 B) by intracellular acidification (Wang et al., 1992; Tsai et al., 1995).
- As described in section 1.2.3, aldosterone stimulates reabsorption of Na^+ and secretion of K^+ in the cortical collecting duct. Elevation of this hormone during high K^+ diet was hypothesized to increase $K_{ir}1.1$ activity, at least in part, by increasing the intracellular pH via stimulation of Na^+/H^+ -antiport (Oberleithner et al., 1987).
- For effective K^+ secretion in principal cells the activity of basolateral Na^+/K^+ -ATPase must be coupled to the apical K^+ conductance. As an alternative to the mechanism proposed by Wang et al. (1993), activation of Na^+/K^+ -ATPase could enhance Na^+/H^+ -antiport, thereby increasing pH_i and apical $K_{ir}1.1$ channel activity.

- As a more general mechanism, pH-gating of K_{ir} channels could play a protective role in secretory epithelia. To maintain cation balance during K^+ secretion, epithelial cells exchange K^+ with Na $^+$ or H $^+$ (Lang and Rehwald, 1992). Inactivation of K_{ir} channels as a result of excessive uptake of protons would prevent further acidification of these cells.

Although an increase of K^+ secretion at higher luminal K^+ concentrations has been described (Giebisch, 1998), a direct activation of $K_{ir}1.1$ channels by luminal K^+ ions has not yet been taken into account. The discovery of K^+ regulation and its allosteric interaction with pH-gating of $K_{ir}1.1$ channels delivers a molecular explanation for this phenomenon and opens some interesting perspectives. K^+ secretion increases the tubular K^+ concentration, which would activate apical $K_{ir}1.1$ channels. This mechanism therefore might represent a positive feedback control of the K^+ secretion process which would otherwise be negatively affected by the decreased electrochemical driving force that accompanies increased luminal K^+ concentrations. Since the K^+ level in the blood plasma is very stable, the control circuit of K^+ homeostasis must have a high gain. A positive feedback, such as the direct K^+ regulation described here, might increase the overall gain of K^+ control and contribute to the stability of the blood plasma K^+ concentration.

In kidney epithelial cells a drop in intracellular pH induces formation and subsequent elimination of NH_4^+ which helps to control pH_i in these cells (Knepper et al., 1989). However this would require the presence of NH_4^+ -conducting channels which are active at acidic pH_i in the apical membrane. $K_{ir}1.1$ channels were found to conduct NH_4^+ and to be effectively upregulated not only by K^+ but also by NH_4^+ at slightly acidic pH_i (Doi et al., 1996). The idea, that allosteric regulation of $K_{ir}1.1$ channels could affect pH_i or NH_4^+ excretion needs to be investigated in a physiological system.

3.3.3 Differential diagnosis and possible treatment of aBS

 $K_{ir}1.1$ point mutations identified in aBS patients to date are scattered throughout the amino acid sequence. However, when considering the tertiary folding of N- and C-termini as deduced in section 2.3.5, most aBS mutations were found to fall into two categories: (i) mutations in the inner core-region formed by the highly conserved P-loop and (ii) mutations in or close to the determinants of the RKR triad (Fig. 38). In heterologous expression experiments, category (i) mutants did not lead to channel activity (Derst et al., 1997), while most of the category (ii) mutants tested to date encoded functional channels. pH-gating of the latter, however, was shifted to more alkaline pH values compared to $K_{ir}1.1$ wildtype, with backshifts in pK_{app} ranging from about 0.3 pH units ($K_{ir}1.1(L220F)$) to more than 2 pH units ($K_{ir}1.1(R311W)$, $K_{ir}1.1(R311Q)$; see Fig. 39). On the molecular level, these shifts in pH-gating either result from disruption of the RKR triad or may be a consequence of mutation-induced structural

disturbances of this arrangement. Such minor shifts in pK_{app} have been observed for quite a number of amino acid exchanges (see Figs. 29 and 39).

In coexpression experiments $K_{ir}1.1(R311Q)$ behaved as a dominant-negative subunit (Fig. 36). Therefore the corresponding allele should be dominant in terms of the development of aBS. Surprisingly, the carrier of $K_{ir}1.1(R311Q)$ did not show clinical symptoms. It remains to be elucidated if this discrepancy results from non-binominal subunit assembly (compare the relative currents in Fig. 36 resulting from a 1:1 cRNA coexpression with functional $K_{ir}1.1(K80M)$) or overexpression of the functional allele in the kidney.

Under physiological conditions, the mild shift in pK_{app} caused by the $K_{ir}1.1(L220F)$ mutation in one patient was enough to develop the complete clinical phenotype of aBS. The non-functional allele $K_{ir}1.1(R311W)$ he carried on the other chromosome was not sufficient by itself to cause aBS since his mother - the carrier of this allele - was healthy (see "Materials and Methods" section 4.2.4). For this patient, a mild alkalinization of epithelial cells in the distal nephron should be able to restore functional activity of homomeric $K_{ir}1.1(L220F)$ channels and thereby bring this patient close to his mother. Currently such an approach is under clinical trial for a subset of aBS patients with $K_{ir}1.1$ mutations leading to smaller backshifts in pK_{app} .

Taken together, these results suggest that defective pH-gating is a major molecular cause for the pathogenesis of aBS. This may open up a new perspective for differential diagnosis and treatment of at least some of the aBS patients.

3.4 Research perspectives

Several properties reported for $K_{ir}1.1$ channels in kidney cells could not be reproduced for cloned $K_{ir}1.1$ expressed in *Xenopus* oocytes, such as block by intracellular ATP and regulation by PKA (Wang and Giebisch, 1991a, b). In a recent publication it was suggested that $K_{ir}1.1$ couples functionally with CFTR, a chloride channel of the ABC protein family (Ruknudin et al., 1998). Upon coexpression, K^+ currents showed ATP-dependent gating and became sensitive to glibenclamide. Whether $K_{ir}1.1$ channels directly associate with CFTR or other ABC proteins and what the functional consequences of such an interaction may be needs to be investigated in more detail. Apart from the properties described above, direct interaction of $K_{ir}1.1$ with membrane proteins could also be involved in state-dependent oxidation of $K_{ir}1.1$ channels in excised inside-out patches. As a future approach, intracellular portions of $K_{ir}1.1$ subunits (expressed in bacteria and purified) could be used to identify possible binding partners (when coupled to a chromatography column, in a blot overlay assay or using the yeast two-hybrid system).

After the structure of the KcsA core region has been resolved by x-ray crystallography, attention is now focused on the structure of intracellular domains of channel proteins and on rearrangements involved in gating. Clearly these goals are beyond the limitations of site-directed mutagenesis. A first step may be structural analysis of purified N- and C-termini of K_{ir} proteins with multidimensional NMR spectroscopy.

Some aspects of the pathophysiology of aBS remain puzzling. For example, why are carriers of a single dominant-negative $K_{ir}1.1$ allele apparently healthy? Are there compensatory mechanisms at the level of $K_{ir}1.1$ or other ion channels or transporters in the nephron? Coexpression experiments with $K_{ir}1.1$ mutants in the same combinations as found in compound heterozygous patients may help to answer these questions. After all it would also be important to actually measure the intracellular pH (below the apical membrane) in renal epithelial cells under various physiological conditions. Perhaps $K_{ir}1.1$ channels themselves could be used to monitor changes in intracellular pH.

4 Materials and Methods

4.1 Equipment and materials

4.1.1 Laboratory equipment

Molecular biology:

centrifuge Microfuge and Centrifuge 5417C, Eppendorf, Hamburg, Germany

DNA sequencing ALF express sequencing system, Pharmacia Biotech, Uppsala,

Sweden

film processing Fuji RGII X-ray Film Processor, Fuji Photo Film (Europe)

GmbH, Düsseldorf, Germany

Germany; camera E.A.S.Y. 429K and personal computer,

Herolab, Wiesloch, Germany

gel electrophoresis Horizon 58, Life Technologies, Gaithersburg MD, USA; power

supply EPS 600, Pharmacia Biotech, Uppsala, Sweden

PCR machine DNA Thermal Cycler, Perkin Elmer, Norwalk CT, USA

pipettes Pipetman P2 / P20 / P200 / P1000, Gilson Medical Electronics,

Villiers-le-Bel, France

shakers MS1 IKA Works Inc., Wilmington NC, USA, and Thermomixer

5436 Eppendorf, Hamburg, Germany

temperature-controlled bath GFL 1003, Gesellschaft für Labortechnik mbH, Burgwedel,

Germany

Electrophysiology:

acquisition software Pulse++ (non-commercial software), developed by U. Rexhausen,

Department of Physiology II, University of Tübingen, Germany

balance Mettler AE 163, Mettler Waagen, Giessen, Switzerland

binocular Wild M3C, Leica, Bensheim, Germany; light source: Leica IL

1500 electronic, Schott-Geräte, Germany

chart recorder Kipp&Zonen BD 41, Kipp&Zonen, Netherlands

computational software IgorPro, Wavemetrics, USA

computer Macintosh PowerPC 8500/120, Tübingen, Germany

digitization board ITC16, HEKA electronics, Lamprecht, Germany

electrode puller DMZ-Universal Puller, Zeitz-Instrumente, Augsburg, Germany

incubator WTB Binder, Tuttlingen, Germany

inverted microscope Axioskop, Zeiss, Germany

magnetic stirrer IKA Combimag RET, Janke&Kunkel, Staufen, Germany

microforge microscope ID 03 Carl Zeiss, Germany; power supply

Luigs&Neumann, Ratingen, Germany

micromanipulator, manual Leitz, Wetzlar, Germany

micromanipulator, programmable

Eppendorf, Hamburg, Germany

oscilloscope HM 1007, Hameg, Frankfurt, Germany

patch clamp amplifier EPC9 amplifier, HEKA electronics, Lamprecht, Germany

pH meter pH-Meter CG 840, Schott-Geräte, Hofheim, Germany

pH microelectrode amplifier built by U. Schüler, Department of Physiology II, University of

Tübingen, Germany

precision forceps (size 3 and

size 5)

Dumont, Basel, Switzerland

pump WISA, Wuppertal, Germany

two-electrode voltage-clamp

amplifier

TurboTec 01C amplifier, npi electronic GmbH, Tamm, Germany

ultrasonic bath Transsonic T420, Elma, Singen, Germany

Immunocytochemistry:

an epifluorescense microscope Zeiss Axioskop FS, Jena, Germany

CCD camera Princeton Instruments, Trenton, USA

microtome Leica, Wetzlar, Germany

4.1.2 Materials

Molecular biology:

filter tips SafeSeal Tips, Biozym Diagnostic, Oldendorf, Germany

glass wool Supelco, Belleforte PA, USA

petri dishes (sterile) Greiner, Frickenhausen, Germany

x-ray films Biomax MR, Eastman Kodak Company, Rochester NY, USA

Electrophysiology:

filter holders FP 030/3 0.2 µm, Renner, Dannstadt, Germany

borosilicate glass capillaries Zinsstag Glasbläserei, Stuttgart, Germany

capillaries with filament GB 200F-8P, Science Products, Hofheim, Germany

loading capillaries transferpettor cups (orange) 10 µl, Brand, Deutschland

microloaders Eppendorf, Hamburg, Germany

Immunocytochemistry:

embedding polymer Technovit 7100, Heraeus Kulzer, Wehrheim, Germany

glass slides Superior, Germany; sealed with Entellan, E. Merck, Darmstadt,

Germany

4.1.3 Chemicals and reagents

Molecular biology:

5'-Cy5 labelled primers Gibco BRL, Life Technologies, Karlsruhe, Germany

agar Roth, Karlsruhe, Germany

agarose electrophoresis grade, Life Technologies, Paisley, Scotland

ampicillin Sigma, St. Louis, USA

cRNA synthesis and DNAse I mMessage-mMachine kit, Ambion, Austin, USA

ethidium bromide Sigma, St. Louis, USA

gel electrophoresis CleanGel DNA Analysis Kit, Pharmacia Biotech, Uppsala, Sweden

LB medium Luria Broth base, Gibco BRL, Life Technologies, Paisley,

Scotland

mineral oil Sigma, St. Louis, USA

nucleotides nucleotide mix, Stratagene Europe, Amsterdam, Netherlands

oocyte expression vector pBF (B. Fakler, unpublished)

PCR buffer Gibco BRL, Life Technologies, Karlsruhe, Germany

plasmid preparation Midi Prep Kit, Quiagen, Hilden, Germany

restriction enzymes Boehringer Mannheim, Germany

RNA standards 0.24-9.5 kb RNA Ladder, Life Technologies, Karlsruhe, Germany

Taq polymerase Gibco BRL, Life Technologies, Karlsruhe, Germany

Electrophysiology

antibiotics stock solution (10.000 U penicillin and 10 mg streptomycin per

ml), Sigma, St. Louis, USA

collagenase type II clostridiopeptidase A, EC 3.4.24.3, Sigma, St. Louis, USA

Cu-phenanthroline Aldrich-Chemie, Steinheim, Germany

DTT, DTNB Sigma, St. Louis, USA

FmocCl Sigma, St. Louis, USA

GSH Sigma, St. Louis, USA

H⁺-selective polymer pH ionophore II, Fluka Chemie, Germany

hexamethyl-disilazane Fluka Chemie, Germany

MgATP, K2ATP Sigma, St. Louis, USA

MTSES Toronto Research Chemicals, North York, Canada

PIP2 Boehringer Mannheim, Germany

SIN-1 Calbiochem, San Diego, USA

Spermine Sigma, St. Louis, USA

Immunocytochemistry

goat serum Gibco BRL, Life Technologies, Karlsruhe, Germany

paraformaldehyde Sigma, St. Louis, USA

primary antibody mouse anti-flag antibody M2, Sigma Kodak, USA

secondary antibody goat anti-mouse Cy3-coupled, Dianova, Hamburg, Germany

4.2 Heterologous expression in *Xenopus* oocytes

4.2.1 cRNA synthesis

Site-directed mutagenesis and construction of K_{ir} chimeras was performed with standard techniques (described in (Fakler et al., 1995)) by Dr. J. Ludwig, Dr. H. Hahn, and S. Weidemann (Department of Physiology II, Eberhard-Karls-University of Tübingen, Germany) and Dr. C. Derst (Institute for Physiology, Philipps-University of Marburg, Germany). K_{ir} mutants were verified by sequencing and subcloned into pBF expression vector. Linearized template DNA was kindly provided for *in-vitro* synthesis of capped cRNA using the mMessage-mMachine kit. RNAse-free water (DEPC-water) was prepared as follows: 50 ml $H_2O_{bidest} + 50$ µl diethyl-pyrocarbonate were mixed, incubated overnight at 37°C and autoclaved. Transcription was carried out according to the manufacturer's instructions:

6 µl template DNA

2 µl 10x transcription buffer

10 µl 2x ribonucleotide mix

2 µl SP6 polymerase (premix)

were mixed and incubated for 1.5 h at 37°C. After that, 1 μ l DNAse I (RNAse free, 2 U/ μ l) was added and the mixture was incubated for 15 min at 37°C. Finally, 80 μ l RNAse free water was added and proteins were extracted with phenol / chloroform (mixture + 1 volume of phenol + 1 volume of chloroform upper aequeous phase washed again with 1 volume of chloroform).

RNA was precipitated from the aequeous phase with ethanol: 3 volumes ethanol and 0.1 volumes 3 M sodium acetate pH 4.8 were mixed and incubated on ice for at least 5 min. After centrifugation (5 min at 14000 rpm, Microfuge) the pellet was washed with 70% ethanol, lyophilized and dissolved in DEPC-water. Aliquots were tested on denaturing agarose gels,

fluorescense signals were quantified using RNA standards (0.24-9.5 kb RNA Ladder) and the RNA concentration was finally adjusted to $0.5 \mu g/\mu l$ after repeated precipitation.

4.2.2 Preparation and injection of oocytes

Xenopus oocytes were surgically removed from adult females and manually dissected. About 50 nl of a solution containing cRNA (concentration 50 μ g/ μ l or less, diluted with RNAse-free water) was injected into Dumont stage VI oocytes. Two days later, oocytes were treated 15 min with collagenase type II (Clostridiopeptidase A), washed several times and incubated at 19 °C for at least 2 more days. Through all these procedures, oocytes were kept in OR-2 culture medium.

For two-electrode voltage-clamp experiments only the follicular cell layer was removed using precision forceps. For patch-clamp experiments, the respective oocyte was placed in hypertonic solution (200 mM K⁺ aspartate, pH 7.2 adjusted with KOH), allowed to shrink and all extracellular layers were removed with precision forceps. Oocytes were then placed in the bath chamber and allowed to adhere to the glass bottom (at least 5 min).

4.2.3 Culture media

OR-2 (oocyte culture medium):

NaCl	82.5 mM		
KCl	2.5 mM		
Na_2HPO_4	1.0 mM		
HEPES	5.0 mM		
PVP	0.5 g/l		
$MgCl_2$	1.0 mM		
CaCl ₂	1.0 mM		

pH 7.3 was adjusted with NaOH, and antibiotics (premixed stock solution containing 10.000 U penicillin and 10 mg streptomycin per ml) were added (10 ml per liter OR-2) to prevent bacterial contamination.

4.2.4 K_{ir}1.1 mutations identified in aBS patients

Apart from the $K_{ir}1.1$ -aBS mutations published, six missense mutations were tested which had only recently been identified in three (compound heterozygous) patients (Jeck, N., and Konrad, M., Department of Pediatrics, Philipps-University Marburg, Germany):

- patient 1: $K_{ir}1.1(R311W)$ and $K_{ir}1.1(L220F)$
- patient 2: K_{ir}1.1(R311Q) and K_{ir}1.1(F325C)
- patient 3: $K_{ir}1.1(C49Y)$ and $K_{ir}1.1(I51T)$

All these patients displayed a clinical course indicative of defective ROMK function including marked polyhydramnios with premature birth, marked hyperkalemia in the postnatal period and hypokalemia later on, severe salt wasting and hyperreninemic hyperaldosteronism.

Mutation analysis was performed as follows (Derst et al., 1998; Konrad et al., 1999). Genomic DNA was extracted from peripheral leukocytes isolated from the patients' blood. Aberrant band patterns for the *romk* gene were sought by means of single-strand conformation polymorphism analysis (SSCP) using primers described previously (Karolyi et al., 1997). PCR was performed in a 20 μl volume containing 50 ng of genomic DNA, 1.5 mM MgCl₂, 5 mM Tris (pH 8.3), 50 mM KCl, 10 pmol of each primer and 1.0 U of Taq polymerase. After an initial step at 94 °C for 5 min, PCR was conducted for 30 cycles with denaturation at 94 °C for 45 s, annealing at 55 °C for 30 s and extension at 72 °C for 45 s. The reaction was completed with a final elongation step at 72 °C for 10 min. Amplified products were separated using the CleanGel DNA Analysis Kit with the Multiphor II electrophoresis system. Migration was performed at 18 W constant power at 15 °C for 1 h. The band patterns were visualized by silver staining (Derst et al., 1998). Direct sequencing was performed after reamplification of the remaining PCR product using 5'-Cy5 labelled primers on an ALF express sequencing system following the protocols provided by the manufacturer.

4.3 Immunocytochemistry

Immunostainings were performed in collaboration with Dr. K. Wild using the following procedure (Wild, 1999). After removal of the vitelline membrane oocytes were fixed overnight at -20 °C in fixative (Dent et al., 1989). The fixation reagent was washed out with PBS (137 mM NaCl, 2.7 mM KCl, 8.4 mM Na₂HPO₄·2H₂O, pH 7.2 adjusted with HCl). Then oocytes were incubated with a mouse anti-flag antibody (dilution 1:1000 in PBS + 10% goat serum) at room temperature for 3 h. After washing with PBS, incubation with the secondary

antibody (goat anti-mouse Cy3-coupled, dilution 1:200) was performed either overnight at 4 $^{\circ}$ C or at room temperature for one hour. Stained oocytes were washed 8 times in PBS (for increasing periods of time, 5 min to 1 hour) and post-fixed with 3.7% paraformaldehyde in PBS for one hour. Embedding in Technovit 7100 was carried out according to the manufacturer's instructions. Embedded oocytes were cut to 4 μ m sections, covered on glass slides and analyzed with an epifluorescence microscope. Photos were taken with a CCD camera and stored on harddisk.

4.4 Whole-cell recording

4.4.1 Two-electrode voltage-clamp

Electrophysiological recordings were performed 3 to 7 days after injection. For two-microelectrode recordings current and voltage electrodes were pulled from thick-walled borosilicate glass, had resistances between 0.1 and 0.5 M and were filled with 3 M KCl. Currents were recorded with a TurboTec 01C amplifier, digitized at 0.1 kHz and stored on harddisk. The bath chamber was designed as a narrow canal to achieve complete solution exchange in less than 10 s.

Table 2. Composition of bath solutions used for two-electrode voltage-clamp experiments; all concentrations are given in mM. Intracellular acidification was achieved by replacing KCl in 90K with equimolar concentrations of KHCO₃ (90KHCO₃). The pH value was adjusted to 7.2 by titration with HCl.

Bath solution	2.5K	0K	20K	90K
NaCl	115	117.5	97.5	27.5
KCl	2.5	-	20	90
HEPES	10	10	10	10
CaCl ₂	1.8	1.8	1.8	1.8

4.4.2 Recording of intracellular pH (pH_i)

pH-sensitive electrodes were made and calibrated as described (Tsai et al., 1995). Microelectrodes were filled with hexamethyl-disilazane and baked for 10 min at 200 °C. Tips of the silanized electrodes were filled with a H^+ -selective polymer (pH ionophore II). The rest of the electrode was filled with K_{int} pH 7.0. Calibration was carried out in pH 6.5 and pH 7.5 solutions and only electrodes with linear slopes of > 50 mV/pH unit were used for experiments. The intracellular pH was determined during voltage-clamp experiments by measuring the potential of the pH electrode relative to the command voltage. The measured potential was low-pass filtered (cut-off frequency of 0.1 Hz) and stored on harddisk. In pH_i traces such as in Fig. 24 only a single data point per voltage ramp (at 0 mV) is shown.

4.4.3 Voltage protocols and data evaluation

The voltage-command protocol used for all whole-cell measurements consisted of repetitive voltage ramps: hyperpolarization at -120 mV for one second followed by a linear ramp up to +50 mV in 20 s. For the recording in Fig. 2, short voltage ramps (from -100 mV to +100 mV in 5 s) were used. Stored data were analyzed with commercial software on a Macintosh PowerPC. Activation and deactivation time courses were fitted with monoexponential functions to the current amplitudes at -80 mV:

$$I(t) = I_{\text{max}} \cdot e^{(\text{-t/})} + I_0 \quad \text{(with $I_{\text{max}} = \text{maximal current, } I_0 = \text{steady-state current, } = \text{(in)activation time constant)}$$

All mean values are given as mean \pm standard deviation of n experiments.

4.5 Patch-clamp experiments

4.5.1 Giant inside-out patch-clamp technique

Giant patch-clamp recordings were performed as described previously (Fakler et al., 1995). Briefly, pipettes were made from thick-walled borosilicate glass and had resistances of 0.3 - 0.6 M (tip diameter of 20 - 30 μ m). Currents were sampled at 1 kHz (unless stated otherwise) and corrected for capacitive transients with an EPC9 amplifier, with analog filter set to 3 kHz (-3 dB). Solutions were applied to the cytoplasmic side of the excised patches via a multi-barrel pipette.

Table 3. Solutions used in giant inside-out patch-clamp experiments. All concentrations are given in mM. pH of bath and pipette solutions were adjusted to pH 7.2 with KOH (NaOH in the case for $Na_{pipette}$). K_{int} was first adjusted to pH 8.0 with KOH and then titrated with KOH or HCl to the pH values indicated.

Solutions	bath	$\mathbf{K}_{pipette}$	Napipette	K _{int}	$\mathbf{NMDG}_{\mathrm{int}}$
NaCl	-	-	120	-	-
KCl	100	120	-	120	-
NMDG	-	-	-	-	120
HEPES	10	10	10	10	10
K ₂ EGTA	10	-	-	10	10
$MgCl_2$	1.44	-	-	-	-
CaCl ₂	-	1.8	1.8	-	-

DTT (100 mM stock solution, pH 7.2 adjusted with KOH) was added to all K_{int} solutions at a final concentration of 0.1 mM prior to the experiments unless stated otherwise. All other compounds were prepared as stock solutions and applied in the concentrations indicated.

- Spermine: Stock solution of 100 mM, stored at -20°C
- MgATP, K₂ATP: Stock solution of 0.5 M K₂ATP (pH 6.0 adjusted with KOH). Added together with MgCl₂ to the applied K_{int} solutions to yield the free concentrations indicated. Total and free concentrations of Mg²⁺ and ATP were calculated using the Fabiato program (Fabiato, 1988).
- PIP2: added to solutions at the final concentration and sonified for at least 30 min. These solutions were used for one day.

All experiments were performed at room temperature (approximately 23 °C).

4.5.2 Chemical modification of cysteine

For DTT and DTNB, stock solutions (100 mM) were made and stored at -20 °C; the final dilutions were used for about 8 hours. MTSES was freshly prepared at the final concentration (100 μ M) prior to each experiment and used within 20 min. Glutathione was dissolved in K_{int} solution at a concentration of 5 mM. The NO-donor SIN-1 was prepared as a 100 mM stock solution on ice and used for 4 hours. S-Nitrosocysteine was freshly prepared according to Broillet and Firestein (1997) from equimolar amounts of NaNO₂ and cysteine. Stock solutions (100 mM) were kept on ice and used for 1 hour.

4.5.3 Chemical modification of lysine

FmocCl was chosen due to its high reactivity and selectivity for unprotonated amino groups. Water-free dioxane was used as solvent for the stock solution (100 mM). Three microliters were added to 3 ml of modification buffer no later than 1 min prior to each experiment to yield a final concentration of 0.1 mM FmocCl and 0.1% dioxane.

Modification buffer (MOD):

 $\begin{array}{ccc} \text{KCl} & 100 \text{ mM} \\ \text{H}_3 \text{PO}_4 & 10 \text{ mM} \end{array}$

pH was adjusted to 7.5 with KOH.

4.5.4 Voltage protocols and data evaluation

For evaluation of pH dependence, the steady-state value for relative inhibition (I_{rel}) at a given pH_i was determined by monoexponential fits to the time course of pH_i-induced current inhibition. This value was normalized to the maximum current, which was reached at pH_i = pK_{app} + 1 pH unit in the case of most pH-gated K_{ir} channels. When the apparent pK was higher than 8.0, the current amplitude at pH_i 8.8 was chosen as reference value because of the limited patch stability at higher pH_i.

To obtain pK_{app} and the Hill coefficient for the pH dependence, I(X) was fitted to a modified Hill equation:

$$I_{rel} / I_0 = 1 / \{1 + (X / K_{0.5})^{N}\}$$
; with $K_D = K_{0.5}^{N}$

 $(X = concentration of intracellular \ H^+ ions ([H^+]_i), \ K_{0.5} \ is \ [H^+]_i \ at half maximal inhibition (= 10^{-pK_{app}}), \ N = Hill coefficient and \ I_0 \ is the asymptotic value of \ I_{rel}).$

Fractional recovery of oxidized channels by DTT (rel. recovery) was calculated as follows:

$$rel. \; recovery = [\; I_{pH_{8.0}}(+DTT) \; - \; I_{pH_{8.0}}(after) \;] \; / \; [I_{pH_{8.0}}(before) \; - \; I_{pH_{8.0}}(before) \;] \; / \; [I_{pH_{8.0}}(before) \; - \; I_{pH_{8.0}}(before) \;] \; / \; [I_{pH_{8.0}}(before) \; - \; I_{pH_{8.0}}(before) \;]$$

with $I_{pH_{8.0}}(+DTT)$ steady-state current amplitude at pH_i 8.0 in the presence of DTT, $I_{pH_{8.0}}(after)$ current amplitude at pH_i 8.0 after acidification, $I_{pH_{8.0}}(before)$ current amplitude at pH_i 8.0 before acidification.

5 References

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6 Summary

ROMK ($K_{ir}1.1$), a member of the family of inwardly rectifying (K_{ir}) potassium channels, forms the channel responsible for K^+ secretion in the distal tubulus of the kidney. K^+ homeostasis is strictly controlled by the intracellular pH in a way that $K_{ir}1.1$ channels close in response to a decrease in pH_i. In this study it was shown that pH-gating is associated with conformational changes in the $K_{ir}1.1$ protein. Firstly, conformational states of open and closed $K_{ir}1.1$ channels were probed intracellularly by cysteine modification. Irreversible modification by sulfhydryl reagents was observed for $K_{ir}1.1$ channels only in the closed but not in the open state. Mutagenesis of intracellular cysteines revealed cysteine 49 in the N- and cysteine 308 in the C-terminus as targets for state-dependent modification. Secondly, $K_{ir}1.1$ channels have been shown to be allosterically regulated by extracellular K^+ and intracellular pH. At alkaline pH_i, $K_{ir}1.1$ channels remained active in the absence of extracellular K^+ but failed to recover from pH-induced inactivation. Dependence on extracellular K^+ was abolished in a pH-gated $K_{ir}1.1$ chimera ($K_{ir}1.1(K_{ir}2.1p)$), where the P-loop had been replaced by the corresponding sequence of $K_{ir}2.1$. These results indicate that pH-gating of $K_{ir}1.1$ channels involves conformational changes in both N- and C-termini as well as in extracellular parts of the pore.

Among all inward-rectifiers tested, only $K_{ir}1.1$ and $K_{ir}4.1$ displayed sensitivity to pH_i in the physiological range ($pK_{app} = 6.8$ / Hill coefficient of 2.9 and $pK_{app} = 6.1$ / Hill coefficient of 2.3, respectively). By sequence alignment and site-directed mutagenesis a lysine residue N-terminal to M1 (K80) was identified as the pH_i sensor for pH-gating. Replacing lysine 80 by a neutral amino acid like methionine or modification with an aminogroup-specific agent removed pH-gating of $K_{ir}1.1$ channels. Conversely, introduction of lysine at the pre-M1 site conferred pH-gating to other K_{ir} channels. These results implied that titration of lysine 80 was shifted by more than 3 pH units compared to the free amino acid in all $K_{ir}1.1$ channels tested. By neutralization of conserved positive charges in $K_{ir}1.1$ it was found that the anomalous titration was due to electrostatic interactions within an intrasubunit triad of basic amino acid residues formed by lysine 80, arginine 41 in the N- and arginine 311 in the C-terminus of $K_{ir}1.1$. These results provided the first rough network for the tertiary folding of intracellular domains of K_{ir} channels.

Disruption of the RKR triad resulted in defective ROMK function and is found in families with antenatal Bartter-syndrome (aBS). Moreover, the majority of known $K_{ir}1.1$ -aBS missense mutations is found to be located close to the triad in a structural model deduced from the findings reported here. Analysis of these mutants revealed titration shifts ranging from +0.3 to +2.2 pH units compared to $K_{ir}1.1$ wildtype. This suggests that structural disturbance of the triad in $K_{ir}1.1$ is a major molecular cause for the pathogenesis of aBS.

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